

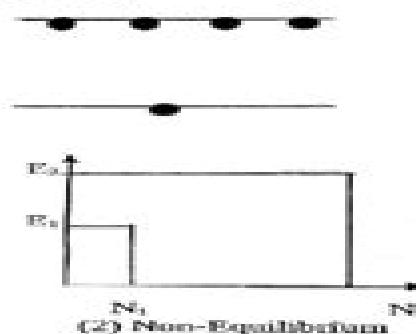
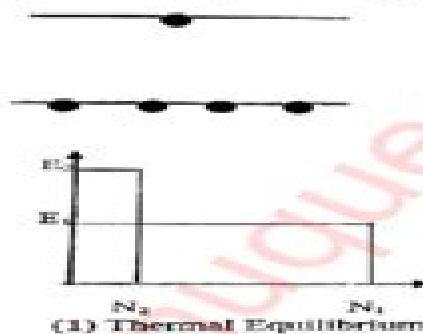
Q1] a) Explain following terms i) spontaneous emission ii) population inversion. (3)

ANS : (i) Spontaneous emission : -

After the life time of the state is over, the atom in excited state gets de-excited to lower energy level on its own. This lower energy state may be either ground state or still one of the excited states but having lower energy level. In the process, a photon is emitted. In this emission process, Where the atoms spontaneously goes to a lower energy state through the emission of a photon is called Spontaneous Emission.

(ii) Population Inversion: -

In thermal-equilibrium the population N_1 of lower energy level is very large as compared to population N_2 of upper energy level, i.e. $N_1 \gg N_2$. The non-equilibrium state in which the population N_2 of the upper energy level exceeds to a large extent the population N_1 of the lower energy level is known as the state of population inversion. i.e. $N_1 \ll N_2$.



Q1] b) What is nanotechnology ? what is its significance ? (3)

ANS : Nanotechnology is the design, characterization, production and application of structure, devices and system. The work is done by controlling

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