

# Chemistry Study Guide

Periodic Table of the Elements																		
H	Li	Be																
Na	Mg																	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
Fr	Ra	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo		
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Ac	Th	Pa	U	Ng	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

## Objectives:

To understand the structure of an atom, and the nature + properties of subatomic particles.



## Matter:

Anything that has mass and occupies space.

Examples: air, gases, liquids, solidz!

## Atoms:

Atoms are the smallest, indivisible particle.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protonz (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles.

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p <sup>+</sup>	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e <sup>-</sup>	-0	-1	Orbit/Shell

# Pearson Study Guide For Chemistry

**Paula Yurkanis Bruice**

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Study Guide and Solutions Manual for Essential Organic Chemistry Paula Yurkanis Bruice,2015-03-04 This Study Guide Solution Manual contains learning objectives chapter summaries and outlines as well as examples self tests and concept questions as well as complete step by step solutions to selected problems

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REHS/RS Study Guide: A Guide for Environmental Health Responsibilities and Competencies (5th edition) National Environmental Health Association

(NEHA),2024-01-01 The REHS RS Study Guide reflects the most recent changes and advancements in environmental health technologies and theories Incorporating the insights of 29 subject matter experts from across academia industry and the regulatory community paired with references from over 30 scholarly resources this essential reference is intended to help those seeking to obtain the NEHA Registered Environmental Health Specialist Registered Sanitarian Credential Table of Contents General Environmental Health Statutes and Regulations Food Protection Potable Water Wastewater Solid and Hazardous Waste Hazardous Materials Zoonoses Vectors Pests and Poisonous Plants Radiation Protection Occupational Safety and Health Air Quality and Environmental Noise Housing Sanitation and Safety Institutions and Licensed

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The Study Guide includes learning goals an overview a review section with worked examples and self tests with answers

**Study Guide Chemistry for Changing Times** Richard Jones,John W. Hill,Doris K. Kolb,2006-08 This Study Guide was written specifically to assist you with Chemistry for Changing Times 11th Edition by presenting in condensed form the major concepts theories facts and applications found in the text Every chapter is keyed to the main text and is presented in six sections Key Terms correspond to bold faced terms in the text and represent key expressions in the language of chemistry

Chapter Summaries provide an overview of material to be covered and an outline that can be tailored and annotated with lecture material Chapter Objectives alert you to essential concepts and principles covered in the chapter and serve as checkpoints when you study for exams Discussion food for thought along with common sense commentary about chemistry Examples Problems with Additional Problems modeled on the text problems these examples will help you sharpen your problem solving skills Self Test and Answers practice exams that are designed for self assessment and test preparation Book jacket [Study Guide for Introductory Chemistry](#) Nivaldo Tro,Donna G. Friedman,2017-07-09 This Study Guide was written specifically to assist students using the 6th edition of Introductory Chemistry It presents the major concepts theories and applications discussed in the text in a comprehensive and accessible manner for students It contains learning objectives chapter summaries and outlines as well as examples self tests and concept questions

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