

## Principles of Water

### Multiple Choice

Write the letter of the best answer on the blank provided.

**B**

1. Water is a compound of hydrogen and oxygen formed when: (167)  
A. one hydrogen atom combines with one oxygen atom.  
B. two hydrogen atoms combine with one oxygen atom.  
C. two oxygen atoms combine with one hydrogen atom.  
D. two hydrogen atoms combine with two oxygen atoms.

**C**

2. How much is ordinary fresh water considered to weigh for fire protection purposes? (167)  
A. 5.25 lb/gal (0.63 kg/L)  
B. 7.26 lb/gal (0.87 kg/L)  
C. 8.3 lb/gal (1 kg/L)  
D. 12.8 lb/gal (1.5 kg/L)

**D**

3. When does water exist in a liquid state? (168)  
A. Between 0°F and 100°F (-18°C and 38°C)  
B. Between 16°F and 110°F (-9°C and 43°C)  
C. Between 20°F and 200°F (-7°C and 93°C)  
D. Between 32°F and 212°F (0°C and 100°C)

**D**

4. Water may be used to smother fires in which type of combustible liquid? (168)  
A. Combustible liquids that have no specific gravity  
B. Combustible liquids whose specific gravity is equal to 1  
C. Combustible liquids whose specific gravity is lower than 1  
D. Combustible liquids whose specific gravity is higher than 1

**B**

5. Compared to other common extinguishing agents, water has: (169)  
A. a lower heat-absorbing capacity.  
B. a greater heat-absorbing capacity.  
C. an almost identical heat-absorbing capacity.  
D. an initially greater but then lower heat-absorbing capacity.

**C**

6. When water is converted to steam at 212°F (100°C), it occupies approximately: (169)  
A. 200 times its original volume.  
B. 700 times its original volume.  
C. 1,700 times its original volume.  
D. 3,500 times its original volume.

# Mechanical Concepts Water Study Guide

**Michael Brown**

## **Mechanical Concepts Water Study Guide:**

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## **Table of Contents Mechanical Concepts Water Study Guide**

1. Understanding the eBook Mechanical Concepts Water Study Guide
  - The Rise of Digital Reading Mechanical Concepts Water Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Mechanical Concepts Water Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mechanical Concepts Water Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mechanical Concepts Water Study Guide
  - Personalized Recommendations
  - Mechanical Concepts Water Study Guide User Reviews and Ratings
  - Mechanical Concepts Water Study Guide and Bestseller Lists
5. Accessing Mechanical Concepts Water Study Guide Free and Paid eBooks
  - Mechanical Concepts Water Study Guide Public Domain eBooks
  - Mechanical Concepts Water Study Guide eBook Subscription Services
  - Mechanical Concepts Water Study Guide Budget-Friendly Options
6. Navigating Mechanical Concepts Water Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- Mechanical Concepts Water Study Guide Compatibility with Devices
- Mechanical Concepts Water Study Guide Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Mechanical Concepts Water Study Guide
- Highlighting and Note-Taking Mechanical Concepts Water Study Guide
- Interactive Elements Mechanical Concepts Water Study Guide

8. Staying Engaged with Mechanical Concepts Water Study Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Mechanical Concepts Water Study Guide

9. Balancing eBooks and Physical Books Mechanical Concepts Water Study Guide

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Mechanical Concepts Water Study Guide

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Mechanical Concepts Water Study Guide

- Setting Reading Goals Mechanical Concepts Water Study Guide
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Mechanical Concepts Water Study Guide

- Fact-Checking eBook Content of Mechanical Concepts Water Study Guide
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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