

MCAT Chemistry – Medical College Admission Test Preparation

→ AROMATIC COMPOUNDS

CHAPTER # 24

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1. Benzene is soluble in

(a) alcohol	(b) petrol
(c) ether	(d) ✓ all of these

2. Benzene is insoluble in

(a) ✓ water	(b) alcohol
(c) petrol	(d) ether

3. Chlorination of benzene is:

(a) ✓ An electrophilic substitution reaction	(b) A nucleophilic substitution reaction
(c) A free-radical reaction	(d) An elimination reaction

4. The total number of sigma bonds in benzene are:

(a) 6	(b) ✓ 12
(c) 18	(d) 3

5. The Pi (π) molecular orbital in benzene contains:

(a) 2 electrons	(b) 4 electrons
(c) ✓ 6 electrons	(d) 3 electrons

6. The C – C – C bond angles in benzene are:

(a) ✓ 120°	(b) 90°
(c) 109.28°	(d) 180°

7. The carbon–carbon bond length in benzene is:

(a) longer than a C – C single bond	(b) ✓ longer than a C = C double bond
(c) Shorter than a C = C double bond	(d) Shorter than a C O \equiv C triple bond

8. Two adjacent substituents on benzene ring is said to be:

(a) ✓ Ortho to one another	(b) Meta to one another
(c) Para to one another	(d) None of the above

Mcqs In Aromatic Compounds

AN Whitehead

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in the set and these are all sold as one entity This book provides an up to date and comprehensive account of aromatic chemistry A series of chapters describes the synthesis and reactions of the major functional derivatives of benzene and the more common polycyclic systems The concept of aromaticity and the mechanism of aromatic substitution are discussed as is the use of metals in the synthesis of aromatic compounds Throughout emphasis is placed on mechanisms Worked problems and questions are provided to aid understanding In addition to providing material required by an undergraduate studying chemistry Aromatic Chemistry is also ideal for industrial chemists seeking to update their knowledge of this important aspect of chemistry Ideal for the needs of undergraduate chemistry students Tutorial Chemistry Texts is a major new series consisting of short single topic or modular texts concentrating on the fundamental areas of chemistry taught in undergraduate science courses Each book provides a concise account of the basic principles underlying a given subject embodying an independent learning philosophy and including worked examples **International Books in Print ,1986**

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