

A student investigates the evaporation of water. He pours 100cm^3 of water into measuring cylinder B. Measuring cylinder A is kept at 40°C and B is kept at 80°C in the same part of the laboratory. Fig. 2.1 shows the two measuring cylinders after 3 days.



Fig. 2.1

- (a) (i) Using ideas about molecules, explain why more water evaporates when the water is kept at 80°C .

.....

 [2]

- (ii) Apart from an increase in temperature, state one change that causes water to evaporate faster.

.....
 [1]

- (b) The specific latent heat of vaporisation of water is 2300J/g and the density of water is 1.0g/cm^3 . During the three days, the water level in B drops from the 100cm^3 mark to the level shown in Fig. 2.1. Calculate the energy used to evaporate water from B during the three days.

energy = [2]

O Level Physics Past Papers

John Henry Avery, Michael Nelkon

A red circular graphic with a gradient, appearing as a stylized arrow or a light effect, positioned to the right of the authors' names.

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1. Understanding the eBook O Level Physics Past Papers
 - The Rise of Digital Reading O Level Physics Past Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying O Level Physics Past Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an O Level Physics Past Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from O Level Physics Past Papers
 - Personalized Recommendations
 - O Level Physics Past Papers User Reviews and Ratings
 - O Level Physics Past Papers and Bestseller Lists
5. Accessing O Level Physics Past Papers Free and Paid eBooks
 - O Level Physics Past Papers Public Domain eBooks
 - O Level Physics Past Papers eBook Subscription Services
 - O Level Physics Past Papers Budget-Friendly Options
6. Navigating O Level Physics Past Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - O Level Physics Past Papers Compatibility with Devices
 - O Level Physics Past Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of O Level Physics Past Papers
 - Highlighting and Note-Taking O Level Physics Past Papers
 - Interactive Elements O Level Physics Past Papers
8. Staying Engaged with O Level Physics Past Papers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers O Level Physics Past Papers
9. Balancing eBooks and Physical Books O Level Physics Past Papers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection O Level Physics Past Papers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine O Level Physics Past Papers
- Setting Reading Goals O Level Physics Past Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of O Level Physics Past Papers
- Fact-Checking eBook Content of O Level Physics Past Papers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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