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web nuclear reactions are going on all around us using correctly balanced equations is important whetting to understand nuclear reactions all equations need to be balance to conform to two conservation

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web the first two equations have been balanced for you alpha particle α ${}^4_2\text{He}$ beta particle β gamma ray γ proton p neutron n ${}^1_0\text{n}$ positron β^+ ${}^0_{+1}\text{e}$

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web we can write a balanced nuclear equation for this reaction as let's do a practice example identify the product x by writing a balanced nuclear equation ${}^{137}_{55}\text{Cs} \rightarrow {}^{137}_{56}\text{Ba} + x$ because the mass number is the same on both sides of the equation x must have a

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web identify the missing particle in the following nuclear reaction ${}^{235}_{92}\text{U} + {}^1_0\text{n} \rightarrow {}^{137}_{52}\text{Te} + {}^{96}_{40}\text{Zr} + {}^1_0\text{n}$

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web answer 4 correct in the reactants and products must be the same part b what is the value of in the following nuclear reaction ${}^{65}_{29}\text{Cu} + {}^1_0\text{n} \rightarrow {}^{65}_{28}\text{Ni} + {}^1_1\text{H}$ express your answer as an integer hint 1 how to approach the problem use rule 2 the total of the subscripts atomic numbers mathematically this can be written as answer

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