

Introduction to Chemistry: Chapter 1 Outline

1.1 The Scope of Chemistry

A. What is Chemistry?

1. **Matter**- anything that has mass and takes up space
- You don't have to be able to see something for it to be considered as matter.
- Light is an example of something that does **NOT** have matter
2. **Chemistry**- Study of the composition of matter and the changes matter undergoes
- Chemistry affects all aspects of life and most natural events because all living and nonliving things are made of matter
- Chemistry is known as a central science because it is useful to understanding other sciences.
- Ozone (O₃) is a molecule that absorbs ultraviolet light
- Chlorofluorocarbons-
3. **Mass** - a measurement that reflects the amount of matter
4. **Weight** - is a measurement that reflects the pull of gravity on a given mass
- Mass and weight are different because your weight changes while on the moon

B. Areas of Study

- Five traditional areas of study are organic chemistry, inorganic chemistry, biochemistry, analytical chemistry, and physical chemistry.
- Most chemicals found in organism contain carbon.
 1. **Organic Chemistry**- study of all chemicals containing carbon
 2. **Inorganic Chemistry**- study of chemicals not containing carbon.
- Many inorganic chemicals are found in non-living things.
- 3. **Biochemistry**- Study of processes that take place in living organisms
- 4. **Analytical Chemistry**- Area of study that focuses on the composition of matter (Analyze)
- 5. **Physical Chemistry**- Area that deals with mechanism, rate and energy transfer that occurs when matter undergoes a change. Study of molecular shape and function.
- 6. **Pure Chemistry**- Pursuit of Chemical knowledge for its own use.
- 7. **Applied Chemistry**- research directed toward a practical goal or application.
- An example of applied could be a chemist looking to find a cure for an illness.
- Chemist don't expect for an immediate use for this knowledge, but most chemist research that's designed to answer a specific question.
- Pure and Applied chemistry are linked. Pure research can lead to an application while an application can exist before the research is done to explain how it works.

Two Types of Data:

8. **Qualitative**- data that is descriptive, non-numeric information
9. **Quantitative**- data is a numeric measurement with a unit associated with it.

C. Big Ideas in Chemistry

Big Ideas:

- * Chemistry as the central science- Chemistry overlaps with all of the other sciences.

Outline Of Chemistry A Summary For First Year Students

Heinrich WILL (Professor of Giessen.)

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