

# Introduction to Chemistry: Chapter 1 Outline

## 1.1 The Scope of Chemistry

### A. What is Chemistry?

1. **Matter**- anything that has mass and takes up space
  - You don't have to be able to see something for it to be considered as matter.
  - Light is an example of something that does **NOT** have matter
2. **Chemistry**- Study of the composition of matter and the changes matter undergoes
  - Chemistry affects all aspects of life and most natural events because all living and nonliving things are made of matter
  - Chemistry is known as a central science because it is useful to understanding other sciences.
  - Ozone (O<sub>3</sub>) is a molecule that absorbs ultraviolet light
  - Chlorofluorocarbons-
3. **Mass** - a measurement that reflects the amount of matter
4. **Weight**- is a measurement that reflects the pull of gravity on a given mass
  - Mass and weight are different because your weight changes while on the moon

### B. Areas of Study

- Five traditional areas of study are organic chemistry, inorganic chemistry, biochemistry, analytical chemistry, and physical chemistry.
  - Most chemicals found in organism contain carbon.
    1. **Organic Chemistry**- study of all chemicals containing carbon
    2. **Inorganic Chemistry**- study of chemicals not containing carbon.
  - Many inorganic chemicals are found in non-living things.
    3. **Biochemistry**- Study of processes that take place in living organisms
    4. **Analytical Chemistry**- Area of study that focuses on the composition of matter (Analyze)
    5. **Physical Chemistry**- Area that deals with mechanism, rate and energy transfer that occurs when matter undergoes a change. Study of molecular shape and function.
    6. **Pure Chemistry**- Pursuit of Chemical knowledge for its own use.
    7. **Applied Chemistry**- research directed toward a practical goal or application.
  - An example of applied could be a chemist looking to find a cure for an illness.
  - Chemist don't expect for an immediate use for this knowledge, but most chemist research that's designed to answer a specific question.
  - Pure and Applied chemistry are linked. Pure research can lead to an application while an application can exist before the research is done to explain how it works.
- Two Types of Data:
8. **Qualitative**- data that is descriptive, non-numeric information
  9. **Quantitative**- data is a numeric measurement with a unit associated with it.

### C. Big Ideas in Chemistry

Big Ideas:

- Chemistry as the central science- Chemistry overlaps with all of the other sciences.

# Outline Of Chemistry A Summary For First Year Students

**Heinrich WILL (Professor of Giessen.)**



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