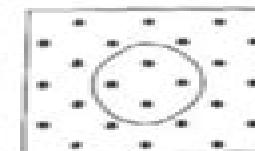
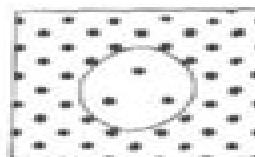
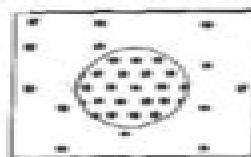


## OSMOSIS STUDY GUIDE

NAME \_\_\_\_\_

Label the pictures below (isotonic, hypertonic, or hypotonic)

Black dots represent solute molecules.



\_\_\_\_\_ tonic means there is a **GREATER** concentration of solute molecules **OUTSIDE** the cell than inside.

\_\_\_\_\_ tonic means there is a **LOWER** concentration of solute molecules **OUTSIDE** the cell than inside.

\_\_\_\_\_ tonic means there is the **SAME** concentration of solute molecules outside the cell as inside.

The pressure inside a plant cell caused by water pushing against the cell wall is called **pressure**.



Cells swell and burst

The **SWELLING AND BURSTING** of animal cells when water enters is called \_\_\_\_\_.

This happens when an animal cell is placed in a \_\_\_\_\_ tonic solution.



Placing plant cells in a **HYPOTONIC** solution causes the osmotic pressure to **increase** **decrease** (Circle ONE)

The **SHRINKING** of plant cells when water leaves so the cell membrane pulls away from the cell wall is called \_\_\_\_\_.



It happens when a plant cell is placed into \_\_\_\_\_ tonic solution.



Cells shrink and shrivel

When water leaves a plant cell, the osmotic pressure will **increase** **decrease** (Circle ONE)

The shrinking of **ANIMAL** cells that are placed in a **HYPERTONIC** solution is called \_\_\_\_\_.

Cells stay the same size when placed in an \_\_\_\_\_ tonic solution because the amount of water leaving the cell is the same as the amount of water entering.

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**Kim Cooper, Kelly Gosnell**

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### **Table of Contents Osmosis Study Guide Answers**

1. Understanding the eBook Osmosis Study Guide Answers
  - The Rise of Digital Reading Osmosis Study Guide Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Osmosis Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Osmosis Study Guide Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Osmosis Study Guide Answers
  - Personalized Recommendations
  - Osmosis Study Guide Answers User Reviews and Ratings
  - Osmosis Study Guide Answers and Bestseller Lists

5. Accessing Osmosis Study Guide Answers Free and Paid eBooks
  - Osmosis Study Guide Answers Public Domain eBooks
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6. Navigating Osmosis Study Guide Answers eBook Formats
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  - Osmosis Study Guide Answers Compatibility with Devices
  - Osmosis Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Osmosis Study Guide Answers
  - Highlighting and Note-Taking Osmosis Study Guide Answers
  - Interactive Elements Osmosis Study Guide Answers
8. Staying Engaged with Osmosis Study Guide Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Osmosis Study Guide Answers
9. Balancing eBooks and Physical Books Osmosis Study Guide Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Osmosis Study Guide Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Osmosis Study Guide Answers
  - Setting Reading Goals Osmosis Study Guide Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Osmosis Study Guide Answers
  - Fact-Checking eBook Content of Osmosis Study Guide Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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