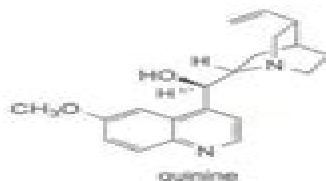


5.65



$$\text{ee} = \frac{[\alpha] \text{ mixture}}{[\alpha] \text{ pure enantiomer}} \times 100\%$$

quinine = A,  
quinine's enantiomer = B

a.  $\frac{-50}{-165} \times 100\% = 30\% \text{ ee}$

b. 30% ee = 30% excess one compound (A)  
remaining 70% = mixture of 2 compounds (30% each A and B)  
Amount of A = 30 = 35 = 85%  
Amount of B = 35%

$\frac{-85}{-165} \times 100\% = 50\% \text{ ee}$

50% ee = 50% excess one compound (A)  
remaining 50% = mixture of 2 compounds (25% each A and B)  
Amount of A = 50 = 25 = 75%  
Amount of B = 25%

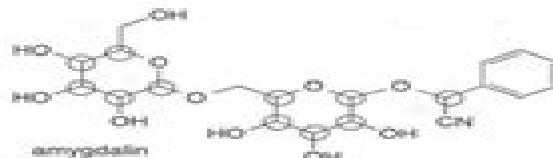
$\frac{-120}{-165} \times 100\% = 73\% \text{ ee}$

73% ee = 73% excess one compound (A)  
remaining 27% = mixture of 2 compounds (13.5% each A and B)  
Amount of A = 73 = 13.5 = 86.5%  
Amount of B = 13.5%

c.  $[\alpha] = +165$   
 $80\% - 20\% = 60\% \text{ ee}$

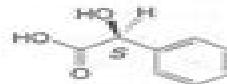
d.  $60\% = \frac{[\alpha] \text{ mixture}}{-165} \times 100\%$   
[α] mixture = -99

5.66



a. The 11 stereogenic centers are circled. Maximum number of stereoisomers =  $2^{11} = 2048$

b. Enantiomers of mandelic acid:



c.  $60\% - 40\% = 20\% \text{ ee}$   
 $20\% = [\alpha] \text{ mixture} - 154 \times 100\%$   
[α] mixture = -31

d.  $\text{ee} = \frac{+50}{+154} \times 100\% = 32\% \text{ ee}$   
32% excess of the S enantiomer

[\alpha] for (S)-mandelic acid = +154

68% of racemic R and S = 34% S and 34% R

S enantiomer: 32% + 34% = 66%  
R enantiomer = 34%

# Organic Chemistry 6th Edition Solution Manual

**Jan William Simek, L. G. Wade**

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### **Table of Contents Organic Chemistry 6th Edition Solution Manual**

1. Understanding the eBook Organic Chemistry 6th Edition Solution Manual
  - The Rise of Digital Reading Organic Chemistry 6th Edition Solution Manual
  - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chemistry 6th Edition Solution Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Organic Chemistry 6th Edition Solution Manual
  - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chemistry 6th Edition Solution Manual
  - Personalized Recommendations
  - Organic Chemistry 6th Edition Solution Manual User Reviews and Ratings
  - Organic Chemistry 6th Edition Solution Manual and Bestseller Lists

5. Accessing Organic Chemistry 6th Edition Solution Manual Free and Paid eBooks
  - Organic Chemistry 6th Edition Solution Manual Public Domain eBooks
  - Organic Chemistry 6th Edition Solution Manual eBook Subscription Services
  - Organic Chemistry 6th Edition Solution Manual Budget-Friendly Options
6. Navigating Organic Chemistry 6th Edition Solution Manual eBook Formats
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  - Organic Chemistry 6th Edition Solution Manual Compatibility with Devices
  - Organic Chemistry 6th Edition Solution Manual Enhanced eBook Features
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  - Adjustable Fonts and Text Sizes of Organic Chemistry 6th Edition Solution Manual
  - Highlighting and Note-Taking Organic Chemistry 6th Edition Solution Manual
  - Interactive Elements Organic Chemistry 6th Edition Solution Manual
8. Staying Engaged with Organic Chemistry 6th Edition Solution Manual
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Organic Chemistry 6th Edition Solution Manual
9. Balancing eBooks and Physical Books Organic Chemistry 6th Edition Solution Manual
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Organic Chemistry 6th Edition Solution Manual
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Organic Chemistry 6th Edition Solution Manual
  - Setting Reading Goals Organic Chemistry 6th Edition Solution Manual
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Organic Chemistry 6th Edition Solution Manual
  - Fact-Checking eBook Content of Organic Chemistry 6th Edition Solution Manual
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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