

Nokia E65 User Guide



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Nokia E65 User Guide

Michael Brown

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we can write a balanced nuclear equation for this reaction as let's do a practice example identify the product x by writing a balanced nuclear equation $^{137}_{55}\text{Cs} + ^{137}_{56}\text{Ba} \rightarrow ^{137}_{56}\text{Ba} + x$ because the mass number is the same on both sides of the equation x must have a

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we identify the missing particle in the following nuclear reaction $^{10}\text{n} + ^{235}\text{U} \rightarrow ^{137}\text{Te} + ^{42}\text{He} + ^{97}\text{Sr} + ^{96}\text{Zr} + ^{40}\text{Zr}$

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