

## 11 • Organic Chemistry

## Organic Nomenclature Practice

Name / Functional Group	Formula	Structural Formula
n-butane	(1) $C_4H_{10}$ (2) $CH_3CH_2CH_2CH_3$	
name & functional group 2-butanol alcohol name only	(1) and (2) $C_4H_9OH$ $CH_3CH_2CHOHCH_3$	
trans-2-butene name only	(1) and (2) $C_4H_8$ $CH_3CH=CHCH_3$	
name only butyne	(1) and (2) $C_4H_6$ $CH_3C\equiv CCH_3$	
name only 1,3-butadiene	(1) and (2) $C_4H_6$ $CH_2=CH=CH=CH_2$	
name only cyclopentane	(1) only $C_5H_{10}$	
name only cis-2-butene	(1) and (2) $C_4H_8$ $CH_3CH=CHCH_3$	
name only trans-2-butene	(1) and (2) $C_4H_8$ $CH_3CH=CHCH_3$	
name & functional group methylethyl ether	(1) and (2) $C_3H_8O$ $CH_3CH_2OCH_3$	

# Nomenclature Problems With Answers

**Oscar Oldberg**



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