



# Pc Build How To

**Mark Dufour**



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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the ph of the solution is then  $ph = -\log 0.100 = 1.000$  ph  $-\log 0.100 = 1.000$  b titrant volume 12.50 ml

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