

True or False Questions

1. The rate of the reaction for the reduction of the Benedict's solution is directly proportional to the concentration of the solution.
☐ **True**
☐ **False**
2. Sodium hydroxide is not required during the production of the Benedict's solution and should be added afterwards.
☐ **True**
☐ **False**
3. In order to make a standard, I can just weigh out over 100 mg of sucrose and dissolve it.
☐ **True**
☐ **False**
4. In order to make standard, standard solutions should be prepared in their containers.
☐ **True**
☐ **False**
5. Heating the solution will speed up the reaction and the rate of the reaction will be faster.
☐ **True**
☐ **False**

Multiple-Choice Questions

6. A standard solution should be:
☐ weighed down the day.
☐ prepared in the water bottle.
☒ **prepared in a separate water container.**
7. The molar mass of sucrose is 342.30 g/mol. I want to make 100 mL of a 0.1 M solution. I need to weigh out 3.4230 g of sucrose.
☐ **3.4230 g of sucrose.**

Miami Dade College Chm 1046 Lab Answers

Nader Rifai



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Strengthening High School Chemistry Education Through Teacher Outreach Programs National Research Council, Division on Earth and Life Studies, Board on Chemical Sciences and Technology, Chemical Sciences Roundtable, 2009-06-15 A strong chemical workforce in the United States will be essential to the ability to address many issues of societal concern in the future including demand for renewable energy more advanced materials and more sophisticated pharmaceuticals High school chemistry teachers have a critical role to play in engaging and supporting the chemical workforce of the future but they must be sufficiently knowledgeable and skilled to produce the levels of scientific literacy that students need to succeed To identify key leverage points for improving high school chemistry education the National Academies Chemical Sciences Roundtable held a public workshop summarized in this volume that brought together representatives from government industry academia scientific societies and foundations involved in outreach programs for high school chemistry teachers Presentations at the workshop which was held in August 2008 addressed the current status of high school chemistry education provided examples of public and private outreach programs for high school chemistry teachers and explored ways to evaluate the success of these outreach programs *Catalog* Florida International University, 1992

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example models the following scenario a salesperson may manage many other salespeople a salesperson is managed by only one salespeople a salesperson can be an agent for many customers

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