

Mystery Powder Lab!

Purpose: One of the employees at our company spilled some of our powders. We now have an unknown substance that we need help identifying. You will first want to learn the properties of each white powder before trying to figure out the mystery combination!

Background: The shape, smell, color, size and how an item reacts are called properties. Some properties are physical and some are chemical. The color, shape, and size are physical properties of an object. Physical properties can be observed without changing the chemical property. Chemical properties include how an object behaves or reacts to something such as an acid. For us to see a chemical property, an item has to actually change chemically. Chemists are the scientists that study the properties of a substance. Both chemical and physical properties are used to identify an item.

In forensic science and chemistry, being able to identify and analyze unknown substances is very important. A police officer might find a white powder at a suspect's house. It's the forensic chemist's job to find out if that white powder is flour, sugar or an illegal drug. With the use of scientific problem solving, the mystery powder can almost always be correctly identified.

Materials:

Corn Starch	Mystery Powder	Magnifying Glass
Salt	Water	Reaction Plate
Plaster of Paris	Vinegar	Pipette Dropper
Baking Soda	Sodium Solution 10%	Scoop

Procedure:

- Use the scoop to put the same amount of corn starch in three of the wells of your reaction plate.
- Observe the powder with your magnifying glass and record your observations in the data table below.

Powder	Magnifying Glass Observations	Water	Vinegar	Scoop
Corn Starch				
Salt				
Plaster of Paris				
Baking Soda				
Mystery Powder				

- Use the pipette to place 10 drops of water into well 1. Stir with your toothpick and then observe how well the powder dissolved in the water. Record this in the data table.
- Gently squeeze 3 drops from the bottle of vinegar into well 2. Observe the color of your powder. Record your observations in the data table.

Vinegar is an indicator for starch, meaning that it shows if something has starch in it. It will stay yellow/brown if the powder doesn't have any starch, but will turn blue/black or purple if the powder has starch in it. If you have extra, vinegar after this experiment, try testing some of the foods in your kitchen to see if they have starch in them!

- Gently squeeze 3 drops from the bottle of vinegar into well 3. Observe what happens and record your observations in the data table.

Vinegar is an excellent tool to react with bases (substances either an acid or a base, two special kinds of chemicals. Examples of acids are lemon juice, vinegar and coffee. Examples of bases are bleach, soap and baking soda. When you mix a stronger acid and base they react to each other to form a new substance).

- Carefully rinse and dry your scoop and reaction plate.
- Repeat steps 1-4 with the salt, plaster of paris, and baking soda.
- Now it is time to figure out your mystery powder! Follow steps 1-4 with it and then use its properties to figure out which two powders combined to make it.

I think the mystery powder is a mixture of _____ and _____ because _____

Questions:

- What properties did all of the powders have in common?
- How can you tell the difference between Cornstarch and Plaster of Paris?
- How can you tell the difference between salt and baking soda?
- How could properties help you tell the difference between each powder?
- What is a physical property? Give an example.
- What is a chemical property? Give an example.

Extension:

Signs that a chemical change has taken place are a change in color, smell, or temperature. Gas bubbles can form and you might hear a sound. Did you see any of these signs when you were testing the mystery powders? Put a star in each box of the data table where a chemical change took place.

Mystery Powder Lab Report

Harold Weisberg

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Christenson, 2019-05-04 Handbook of Student Engagement Interventions Working with Disengaged Students provides an understanding of the factors that contribute to student disengagement methods for identifying students at risk and intervention strategies to increase student engagement With a focus on translating research into best practice the book pulls together the current research on engagement in schools and empowers readers to craft and implement interventions Users will find reviews on evidence based academic behavioral social mental health and community based interventions that will help increase all types of engagement The book looks at ways of reducing suspensions through alternative disciplinary practices the role resiliency can play in student engagement strategies for community and school collaborations in addressing barriers to engagement and what can be learned from students who struggled in school but succeeded later in life It is a hands on resource for educators school psychologists researchers and students looking to gain insight into the research on this topic and the strategies that can be deployed to promote student engagement Presents practical strategies for engagement intervention and assessment Covers early warning signs of disengagement and how to use these signs to

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