

CHAPTER

STUDY GUIDE

The Mole**Section 10.1 Measuring Matter***In your textbook, read about counting particles.**In Column B, rank the quantities from Column A from smallest to largest.***Column A**

- 0.5 mol
- 200
- 5
- 6,000,000,000
- 6.02×10^{23}
- dozen
- four moles
- gross
- pair
- ream

Column B

- 1. _____
- 2. _____
- 3. _____
- 4. _____
- 5. _____
- 6. _____
- 7. _____
- 8. _____
- 9. _____
- 10. _____

*In your textbook, read about converting moles to particles and particles to moles.**In the boxes provided, write the conversion factor that correctly completes each problem.*11.
atoms $1.20 \text{ mol Cu} \times \text{ } = 7.22 \times 10^{23} \text{ Cu}$ 12.
 $\times 10^{-3}$ mol CH₄ $9.25 \times 10^{22} \text{ molecules CH}_4 \times \text{ } = 1.54$ 13.
mol Xe $1.54 \times 10^{22} \text{ atoms Xe} \times \text{ } = 2.56 \times 10^{-3}$ 14.
molecules F₂ $3.01 \text{ mol F}_2 \times \text{ } = 1.81 \times 10^{24}$

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