

- **Thermodynamics** is the study of energy and its transformations.
 - **First Law of Thermodynamics**: Energy cannot be created or destroyed, only transformed.

$$\Delta U = Q - W$$

 where ΔU is the change in internal energy, Q is the heat added to the system, and W is the work done by the system.
 - **Second Law of Thermodynamics**: The entropy of an isolated system always increases over time.

$$\Delta S \geq 0$$

 where ΔS is the change in entropy.



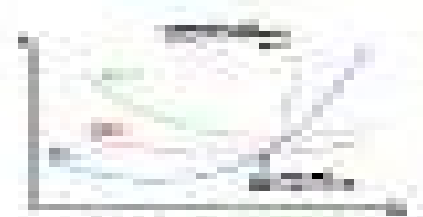
- **Heat Engines** convert heat into work. The efficiency of a heat engine is given by:

$$\eta = \frac{W}{Q_{in}}$$

 where η is the efficiency, W is the work done, and Q_{in} is the heat input.
 - **Carnot Cycle**: A theoretical cycle that operates between two isotherms. The efficiency of a Carnot engine is given by:

$$\eta_{Carnot} = 1 - \frac{T_c}{T_h}$$

 where T_c is the temperature of the cold reservoir and T_h is the temperature of the hot reservoir.



- **Heat Pumps and Refrigerators** transfer heat from a cold reservoir to a hot reservoir. The coefficient of performance (COP) for a heat pump is given by:

$$COP_{HP} = \frac{Q_h}{W}$$

 where COP_{HP} is the coefficient of performance, Q_h is the heat delivered to the hot reservoir, and W is the work input.
 - **Refrigerator**: The COP for a refrigerator is given by:

$$COP_{Ref} = \frac{Q_c}{W}$$

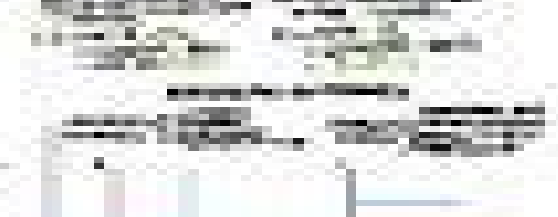
 where COP_{Ref} is the coefficient of performance, Q_c is the heat removed from the cold reservoir, and W is the work input.



- **Entropy** is a measure of the disorder or randomness of a system. The change in entropy for a process is given by:

$$\Delta S = \int \frac{dQ}{T}$$

 where ΔS is the change in entropy, dQ is the infinitesimal amount of heat added, and T is the temperature.
 - **Third Law of Thermodynamics**: The entropy of a perfect crystal at absolute zero is zero.



- **Heat Conduction**: The transfer of heat through a material. The rate of heat conduction is given by:

$$Q = \frac{kA\Delta T}{L}$$

 where Q is the rate of heat conduction, k is the thermal conductivity, A is the cross-sectional area, ΔT is the temperature difference, and L is the length of the material.
 - **Heat Convection**: The transfer of heat by the movement of a fluid. The rate of heat convection is given by:

$$Q = hA\Delta T$$

 where Q is the rate of heat convection, h is the convective heat transfer coefficient, A is the cross-sectional area, and ΔT is the temperature difference.



Microeconomics Final Study Guide

Karl E. Case



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