



THE OUTSIDERS: DISCUSSION GUIDE #3

Encourage critical thinking and support your students' literary development by asking questions during reading. Here are some questions you can ask your readers to engage them in a discussion.



CHAPTERS 9-10

Comprehension:

- What does Ponyboy find before the rumble?
- There are different present signs that figure together in the rumble. How is Tom (Shepard's) gang different from Ponyboy's gang?
- What is Paul's lesson? What is the connection between him and Charly?
- What group wins the rumble?
- What are the effects of the injury Ponyboy suffers in the rumble?
- What happens when Charly learns about Johnny's death?

Text Analysis:

- Ponyboy understands that in his neighbourhood, "growing Pains... would stop up to the old Pains." However, the older think this will happen to him and Charly. Why doesn't Ponyboy see himself as a "pioneer" like Johnny? Is Ponyboy certain that neither the two (Charly) will become Pains?
- A **lesson** is a universal message expressed throughout a work of literature. Who gives the first lesson to Charly and Paul? How they respond to the theme of possibility (between different social classes)?
- An **allusion** is an implied or indirect reference, especially in literature. Johnny's last words to Ponyboy are an allusion: "What does the say, son? What does the mean by this allusion?"

Personal Connections:

- Ponyboy doesn't usually like fighting, and only thinks it's okay in cases of self-defense, but the boys disagreed to participate in the rumble anyway. Can you think of a situation in your life, or real life, where someone didn't want to do something, but did it anyway because they felt like they had to in order to be a good friend? Explain.

Outsiders Study Guide Answer Sheet

Xiaolong Qi

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web they also liaise with local state and federal authorities to ensure the project adheres to the safety laws and regulations

that apply to the construction industry due to the various types of construction projects construction managers can specialize in multiple fields specialization helps them understand the various aspects of a project

spread footing types design advantages and disadvantages civil - May 14 2023

web a combined footing is used when two or more columns in a straight line are borne on a single spread footing the most cost effective option is to use isolated footings for each column combined footings are only used when essential such as building a house

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web jul 17 2023 above 7 points advantages of spread footing are making them ideal choice of different types of construction projects either projects are large or small disadvantages of spread footing spread footing have many advantages but another side it has some disadvantages of spread footing

lab report copper iron stoichiometry 1522 words cram - Jul 02 2022

web the purpose of the copper iron stoichiometry lab was to determine which of the two possible iron sulfate compounds formed when reacting aqueous copper ii sulfate with solid iron metal the techniques used in this lab were weighing by difference quantitative transfer and vacuum filtration

chem 1131 lab report 4 docx stoichiometry of iron copper ii - Apr 30 2022

web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

single replacement stoichiometry lab central bucks school district - Sep 04 2022

web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web lab report stoichiometry copper sulfate and iron the oxidation of trimethylamine by copper ii chloride nov 10 2022 this research was on the irreversible oxidation of trimethylamine by copper ii chloride which takes place at temperatures of about 75 and higher the coordination of trimethylamine by copper ii

lab report on iron stoichiometry 1098 words bartleby - Jan 28 2022

web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

stoichiometry lab the reaction of iron with copper ii sulfate - Oct 05 2022

web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

stoichiometry lab iron and copper ii sulfate tpt - Feb 26 2022

web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

stoichiometry the reaction of iron with copper ii sulfate - Jun 13 2023

web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

lab stoichiometry iron with copper ii sulfate pdf scribd - Nov 06 2022

web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

lab report 4 iron copper ii sulfate studocu - Aug 15 2023

web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

lab 4 lab 4 stoichiometry of iron copper ii sulfate redox - Jan 08 2023

web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out

the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

lab report 4 stoichiometry of iron copper ii sulfate - Feb 09 2023

web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

stoichiometry reaction of iron with copper ii sulfate - Apr 11 2023

web lab 4 stoichiometry the reaction of iron with copper ii sulfate revised 8 19 2009 1 introduction in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated CuSO4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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web stoichiometry the reaction of iron with copper ii sulfate purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g