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example the characteristic part is 2 and the mantissa part is 6452 step 2 to find a corresponding value of the mantissa part use the antilog table using the antilog table find the corresponding value

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sifat sifat gas mulia wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik molekul yg terdiri dari 1 atom ini erat kaitannya dengan struktur elektron oktet dan duplet dari gas mulia

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dec 3 2015 gas mulia dan halogen merupakan unsur kimia yang sifatnya sangat bertolak belakang gas mulia dinamakan mulia karena seperti logam mulia sukar sekali bereaksi sementara halogen halos garam artinya adalah pembentuk garam halogen memiliki 6 bilangan oksidasi sehingga mudah sekali bereaksi dengan unsur lainnya gas mulia noble

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pembuatan gas mulia gas helium mempunyai titik didih yang sangat rendah yaitu 268 8 0c sehingga pemisahan gas helium dari gas alam dilakukan dengan cara pengembunan liquefaction sampai gas alam akan mencair sekitar 156 0c dan gas helium terpisah dari gas alam pembuatan ne ar kr dan xe dilakukan dengan proses destilasi bertingkat

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oct 6 2023 gas mulia ditemukan di dekat golongan halogen dalam tabel periodik karena unsur gas mulia memiliki konfigurasi elektron yang penuh unsur unsur tersebut tidak reaktif dan senyawanya tidak dikenal akibatnya gas gas ini dikenal dengan gas inert

apa itu halogen dalam kimia berikut definisi unsur sifat dan - Nov 21 2022

may 18 2022 dikutip dari buku kimia unsur golongan gas mulia halogen alkali dan alkali tanah yang disusun oleh tantri ambarsari 2020 halogen adalah senyawa yang unsur unsurnya dapat bereaksi dengan logam untuk membentuk garam reaktivitas unsur halogen sangat besar sehingga unsur unsurnya sulit ditemukan di alam