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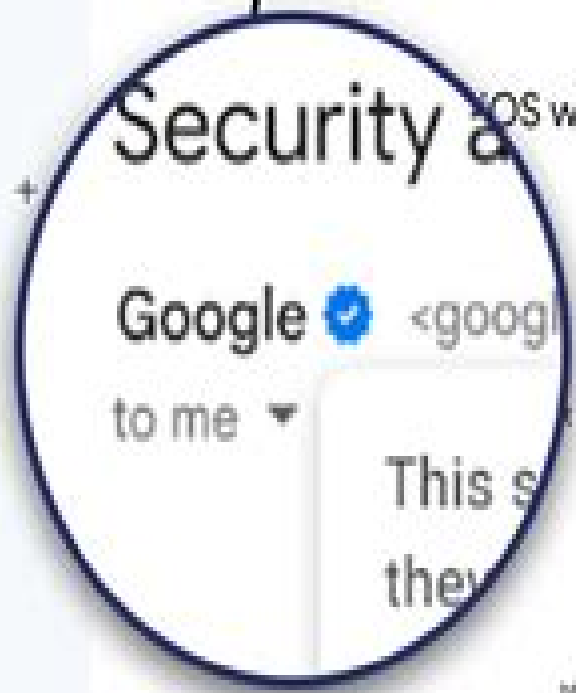
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Hurts Eva Illouz,2013-05-20 Few of us have been spared the agonies of intimate relationships They come in many shapes loving a man or a woman who will not commit to us being heartbroken when we re abandoned by a lover engaging in Sisyphean internet searches coming back lonely from bars parties or blind dates feeling bored in a relationship that is so much less than we had envisaged these are only some of the ways in which the search for love is a difficult and often painful experience Despite the widespread and almost collective character of these experiences our culture insists they are the result of faulty or insufficiently mature psyches For many the Freudian idea that the family designs the pattern of an individual s erotic career has been the main explanation for why and how we fail to find or sustain love Psychoanalysis and popular psychology have succeeded spectacularly in convincing us that individuals bear responsibility for the misery of their romantic and erotic lives The purpose of this book is to change our way of thinking about what is wrong in modern relationships The problem is not dysfunctional childhoods or insufficiently self aware psyches but rather the institutional forces shaping how we love The argument of this book is that the modern romantic experience is shaped by a fundamental transformation in the ecology and architecture of romantic choice The samples from which men and women choose a partner the modes of evaluating prospective partners the very importance of choice and autonomy and what people imagine to be the spectrum of their choices all these aspects of choice have transformed the very core of the will how we want a partner the sense of worth bestowed by relationships and the organization of desire This book does to love what Marx did to commodities it shows that it is shaped by social relations and institutions and that it circulates in a marketplace of unequal actors

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affected by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and
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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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