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web chemlab 16 calorimetry answers 1 the reaction you will be determining the ΔH is the reaction with $\text{NH}_3(\text{aq})$ with $\text{HCl}(\text{aq})$ write the balanced equation for this reaction $\text{NH}_3(\text{aq}) + \text{HCl}(\text{aq}) \rightarrow \text{NH}_4\text{Cl}(\text{aq})$ in addition you will be verifying this by doing two other reactions which add up to the reaction above write balanced equations for these

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