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$$80 \text{ g I}_2 / 1 \text{ mol I}_2 = 1 \text{ mol I}_2$$
$$1 \text{ mol I}_2 \times 133.8 \text{ g/mol} = 133.8 \text{ g I}_2$$
$$133.8 \text{ g I}_2 / 28 \text{ g CO} = 1 \text{ mol CO}$$

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