

11.3. For the excitation transfer process indicated in Fig. 11.1b, write rate equations like (11.3.3) for the populations  $N_A$ ,  $N_B$ , and  $N_{B^*}$ .

We can write the following rate equation for the number density  $N_{A^*}$  of excited atoms  $A^*$  (or molecules) of species A due to the process (11.3.1):

$$\frac{dN_{A^*}}{dt} = -R(N_{A^*}N_B - e^{-\Delta E/k_B T}N_AN_{B^*}). \quad (11.3.3)$$

It is conventional to designate an excited atom or molecule by an asterisk. The exothermic process indicated in Fig. 11.1a is written out as a reaction as follows:



Likewise the endothermic process of Fig. 11.1b is written symbolically as



Figure 11.1 (a) Exothermic and (b) endothermic excitation from atom A to atom B.

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**VM Jensen**

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International Laser Center, Rossiiskai<sup>а</sup> akademii<sup>а</sup> nauk. Institute of Spectroscopy, 2001    Instructor's Solutions Manual for Select Problem Collection to Accompany Fundamentals of Physics David Halliday, Robert Resnick, Jearl Walker, 1998-12-28

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