

Reactions of Amines and Epoxydes

Preparation of Amines

Ammonium hydroxide: $\text{NH}_4^+ + \text{OH}^- \rightarrow \text{NH}_3 + \text{H}_2\text{O}$



Four different types of bonds



There are four types of bonding between atoms in a molecule.

- Ionic bonds are formed by the transfer of electrons from one atom to another (anions and cations) (Opposite charges).



- Covalent bonds are formed by sharing of electrons between two atoms.



Opposite sign and plus (O-H) bonds

Opposite signs are illustrated in oxygen (O-H) bonds.

- Opposite bonds are formed by sharing of electrons.
- O-H bonds are formed by sharing of electrons.
- Very simple bond is a covalent bond.
- Non-polar bonds are formed of two oxygen and two H atoms.



Non-polar bonds are stronger than O-H bonds.

Hydrogen Bonds



Hydrogen bonding



Hydrogen bonding

Hydrogen bonding



Hydrogen bonding

Ammonium hydroxide: $\text{NH}_4^+ + \text{OH}^- \rightarrow \text{NH}_3 + \text{H}_2\text{O}$

Hydrogen bonding

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