

Biochemistry Study Guide

1. Organic molecules
 1. **Macromolecules:** Large complex molecules found only in living organisms.
 2. Define the term **organic molecule.** **Molecules that ave carbon at the center**
 3. Organic molecules are organized around central carbon skeleton.
2. Hydrocarbons
 1. Carbon atoms have four electron in their valence shell, so they can form up to four covalent bonds with other atoms.
 2. Define the term **hydro-carbon.** **Organic molecules consisting entirely of hydrogen and carbon**
 3. Why are hydrocarbons hydrophobic and insoluble in water?
Hydrocarbons are non polar due to slightly differing electronegativities between carbon and hydrogen
 4. Describe the difference between **aliphatic hydrocarbons** and **aromatic hydrocarbons.** Be able to recognize the structural formula of each. **Aliphatic contain a conjugated bond system, while aromatic do not**
3. Organic molecules are important for living organisms, because of their complex structure.
 1. Organic molecules can form isomers. Define the term **isomer.** **A molecule that has the same molecular formula but in different structural formulas**
 2. Organic molecules can form covalent bonds with functional groups. Define the term **functional group.** **A group of atoms that are attached to the carbon skeleton of an organic molecule**
 3. Be able to name the seven functional groups listed in Figure 2.27 on p. 53, and to recognize their structure.
 4. Be able to recognize each of the seven functional groups within a larger organic molecule.
 5. Macromolecules can be very large.
 6. Name the four groups of macromolecules. **Carbohydrates, proteins, nucleic acids, and lipids**
 7. Carbohydrates, proteins, and nucleic acids can be very large, because they are **polymers** that are built from **monomers**.
 8. Monomers are linked together during a **dehydration synthesis.** A water molecule is always a product of a dehydration synthesis.

Organic And Biochemistry Study Guide

Marcia L. Gillette, Wendy Gloffke

Organic And Biochemistry Study Guide:

Introduction to Organic and Biochemistry Peter C. Scott, Morris Hein, 1993-02 Designed for professors who prefer to teach general chemistry topics from one text and organic and biochemistry topics from another this text offers step by step and easy to understand coverage of the important functional groups reactions and macromolecules that are essential for health science students A dynamic full color presentation and numerous applications add to the quality of the presentation Content corresponds to Chapter One and Chapters 21-37 of College Chemistry An Introduction to General Organic and Biochemistry Fifth Edition by the same authors Clarity meticulous accuracy and a step by step approach that students can and do understand have become hallmarks of the Hein authorship This new text is no exception Anticipating student problems before they occur the authors move at a manageable pace offering carefully worked out examples with alternate methods of solution practice problems with answers review of concepts review of key terms and a number of other learning aids to ensure student mastery of important material

General, Organic, and Biochemistry, Study Guide William

Henry Brown, 1983

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Organic and Biological Chemistry, Study Guide John R. Holum, 1996-01-18

Explains recent advances in environmental studies and the molecular basis of life Designed for those in the health care field it focuses solely on organic and biochemistry

Study Guide for Introduction to General, Organic, and Biochemistry William M. Scovell, 2004

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Study Guide for General, Organic, and Biochemistry Wendy Gloffke, Ira Blei, George Odian, 1999-12-20 Expanded from Organic and Biochemistry this text provides a full introduction to general principles for students with little or no exposure to chemistry Topics are presented using familiar everyday examples emphasizing their fundamental importance and personal relevance A CD ROM is packaged with each book and includes animations simulations videos interactive quizzes exercises to help students prepare for exams and a customizable calculator and dynamic periodic table to make problem solving easier The online companion at www.whfreeman.com/chemistry includes CD Web guide chapter objectives Q A interactive exercises key terms and review questions

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time and guides them through the topics in a systematic way Each chapter of the text is covered by an introduction a list of review topics section by section study suggestions and questions a list of key terms and a practice exam with worked out answers *Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry* Susan McMurry,1992

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Fundamentals of General, Organic, and Biological Chemistry, Eighth Edition John McMurry,Susan McMurry,David Ballantine,Carl Hoeger,Virginia Peterson,2016-02-24 This manual provides solutions to all problems in the text It explains in detail how the answers to the in text and end of chapter problems are obtained It also contains chapter summaries study hints and self tests for each chapter

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Rosenberg,2006-10-01 [Study Guide and Solutions Manual for Essentials of General, Organic, and Biochemistry](#) Rachel C. Lum,2014-02-14 **Study Guide and Full Solutions Manual for Fundamentals of General, Organic, and Biological**

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