

Solution

$$r = \frac{V_1}{V_2} = \left(\frac{T_1}{T_2} \right)^{\frac{1}{\gamma-1}} = \left(\frac{646}{323} \right)^{\frac{1}{0.4}} = 5.66 \quad \underline{\underline{\text{Ans}}}$$

$$\begin{aligned} \eta_{\text{Otto}} &= 1 - \frac{1}{r^{\gamma-1}} = 1 - \frac{T_1}{T_2} \\ &= 1 - \frac{323}{646} = 0.5 = 50\% \quad \underline{\underline{\text{Ans}}} \end{aligned}$$

2.3 In an Otto cycle air at 17 °C and 1 bar is compressed adiabatically until the pressure is 15 bar. Heat is added at constant volume until the pressure rises to 40 bar. Calculate the air-standard efficiency, the compression ratio and the mean effective pressure for the cycle. Assume $C_v = 0.717 \text{ kJ/kg K}$ and $R = 8.314 \text{ kJ/kmol K}$.

Solution

Consider the process 1 - 2

$$p_1 V_1^\gamma = p_2 V_2^\gamma$$

$$\begin{aligned} \frac{V_1}{V_2} &= r = \left(\frac{p_2}{p_1} \right)^{\frac{1}{\gamma}} \\ &= \left(\frac{15}{1} \right)^{\frac{1}{1.4}} = 6.91 \end{aligned}$$

$$\begin{aligned} \eta &= 1 - \left(\frac{1}{r} \right)^{\gamma-1} \\ &= 1 - \left(\frac{1}{6.91} \right)^{0.4} = 0.539 = 53.9\% \end{aligned}$$

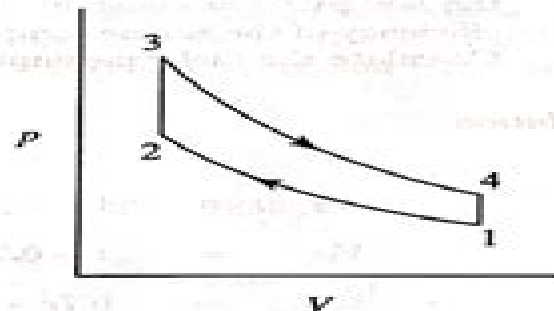
$$T_2 = \frac{p_2 V_2}{p_1 V_1} T_1 = \frac{15}{1} \times \frac{1}{6.91} \times 290 = 629.5 \text{ K}$$

Consider the process 2 - 3

$$T_3 = \frac{p_3 T_2}{p_2} = \frac{40}{15} \times 629.5 = 1678.7$$

$$\begin{aligned} \text{Heat supplied} &= C_v (T_3 - T_2) \\ &= 0.717 \times (1678.7 - 629.5) = 752.3 \text{ kJ/kg} \end{aligned}$$

$$\text{Work done} = \eta \times q_s = 0.539 \times 752.3 = 405.5 \text{ kJ/kg}$$



Otto Cycle Problems And Solutions

Clifford Lane



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A Practical Treatise on the 'Otto' Cycle Gas Engine William Norris, 1896 *Thermodynamics 1 With No Calculations* Edenilson Brandl, Have you ever wanted to understand thermodynamics without getting lost in complex calculations This book provides a clear intuitive approach to one of the most important scientific fields making it accessible to students professionals and curious minds alike By focusing on concepts rather than equations it offers a fresh and engaging way to learn how energy works in the world around us Whether you re a beginner looking for an introduction or an expert seeking a new perspective this book will give you the tools to grasp the principles of thermodynamics with ease With real world examples simple explanations and insightful discussions you will walk away with a deeper appreciation of how thermodynamics shapes our daily lives and the technologies we rely on Don t let complex mathematics stand in the way of understanding thermodynamics Dive into this book and discover how energy heat and entropy govern everything from the engines that power our world to the fundamental processes of nature Get your copy today and unlock the secrets of thermodynamics in the simplest way possible

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