

Chemistry Study Guide

Periodic Table of the Elements

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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

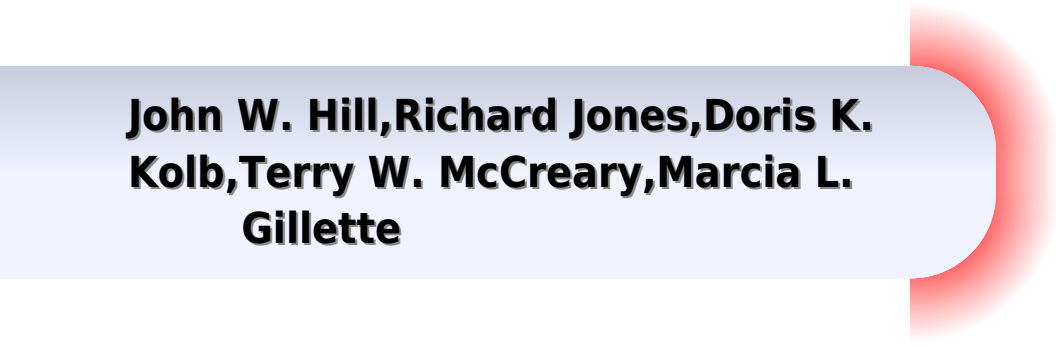
- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Odysseyware Chemistry Study Guide

**John W. Hill, Richard Jones, Doris K.
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Guide for Chemistry Nivaldo J. Tro, Jennifer Shanoski, 2017-01-27 This Study Guide was written specifically to assist students using Structure and Properties It presents the major concepts theories and applications discussed in the text in a comprehensive and accessible manner for students It contains learning objectives chapter summaries and outlines as well as examples self tests and concept questions [Complete Chemistry Study Guide + Flashcards and MCAT Review](#)

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