

Specific Heat Problems – ANSWERS

The last page of this document has Miscellaneous Information that you will need for these problems.

1. How many calories are needed to warm 55.8 g of water at 21.3°C to 73.4°C?

$$\text{Heat} = m \times S. H. \times (T_{\text{final}} - T_{\text{initial}})$$

$$\text{Heat} = 55.8 \text{ g} \times 1.00 \text{ cal/g}^\circ\text{C} \times (73.4^\circ\text{C} - 21.3^\circ\text{C})$$

$$\text{Heat} = 55.8 \text{ g} \times 1.00 \text{ cal/g}^\circ\text{C} \times 52.1^\circ\text{C}$$

$$\text{Heat} = 2907.18 \text{ cal} \quad \text{convert to scientific notation}$$

$$= 2.9\textbf{0}718 \times 10^3 \text{ cal}$$

$$= 2.91 \times 10^3 \text{ cal (3 sig. figs.)}$$

2. How many calories are needed to warm 55.8 g of alcohol from 21.3°C to 73.4°C?

$$\text{Heat} = m \times S. H. \times (T_{\text{final}} - T_{\text{initial}})$$

$$\text{Heat} = 55.8 \text{ g} \times 0.58 \text{ cal/g}^\circ\text{C} \times (73.4^\circ\text{C} - 21.3^\circ\text{C})$$

$$\text{Heat} = 55.8 \text{ g} \times 0.58 \text{ cal/g}^\circ\text{C} \times 52.1^\circ\text{C}$$

$$\text{Heat} = 1686.16 \text{ cal} \quad \text{convert to scientific notation}$$

$$\text{Heat} = 1.6\textbf{8}616 \times 10^3 \text{ cal}$$

$$\text{Heat} = 1.69 \times 10^3 \text{ cal}$$

Organic Chemistry Specific Heat Problems

B Lingard



Organic Chemistry Specific Heat Problems:

Reviewing **Organic Chemistry Specific Heat Problems**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Organic Chemistry Specific Heat Problems**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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