

Practice Final Exam, Chemistry 2220, Organic Chem II

1. Rank the designated protons by chemical shift ( $\delta$ ), highest  $\delta$  first.



- A. IV > III > II > I    B. IV > II > III > I    C. III > IV > II > I    D. III > IV > I > II

2. What are the products, if any, expected from the following reaction?



- $\text{CH}_3\text{OCH}_3 + \text{NaHS}$      $\text{CH}_3\text{SCH}_3 + \text{NaOH}$      $\text{CH}_3\text{O}-\text{H} + \text{CH}_3\text{S}^-\text{Na}^+$     no reaction  
 A.                                  B.                                  C.                                  D.

3. What is the IUPAC name for the following compound?



- A. cis-1,2-cyclopentanediol  
 B. meso-1,2-cyclopentanediol  
 C. (1R,2R)-1,2-cyclopentanediol  
 D. (1R,2S)-1,2-cyclopentanediol

4. Which of the following explains why cyclohexanol has a  $\text{pK}_a$  of 18 and phenol has a  $\text{pK}_a$  of 10?

- A. Phenolate is a stronger base than the conjugate base of cyclohexanol  
 B. The conjugate base of cyclohexanol is resonance stabilized  
 C. The conjugate base of phenol is resonance stabilized  
 D. Phenol is a weaker acid than cyclohexanol



# Organic Chemistry 2 Final Exams

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## **Organic Chemistry 2 Final Exams** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Organic Chemistry 2 Final Exams**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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