

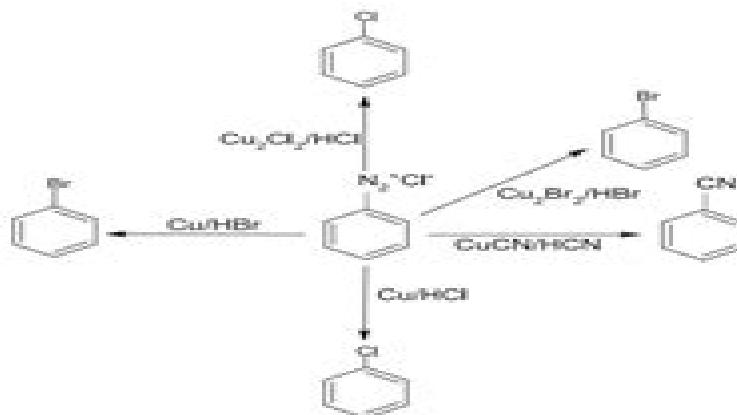
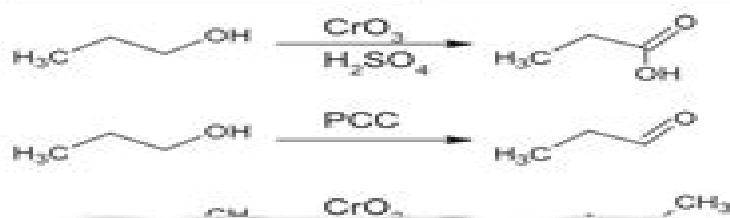
6. CrO_3

Chromium trioxide

- The mixture of chromium trioxide and sulfuric acid dissolved in a mixture of acetone and water is called **Jones reagent**.
- In aqueous acidic solutions, CrO_3 forms chromic acid (H_2CrO_4) which acts as a strong oxidizing agent.
- It is also used to prepare **Pyridinium chlorochromate (PCC)** (**Pyridine + $\text{CrO}_3 + \text{HCl}$**) which is a mild oxidizing agent.

Function -

| Mild oxidizing agent | Strong oxidizing agent |
|---|--|
| 1. converts primary alcohols to Aldehydes | 1. converts primary alcohols to carboxylic acids |
| 2. secondary alcohols to ketones. | 2. secondary alcohols to ketones. |
| 3. NO oxidation of tertiary alcohols | 3. NO oxidation of tertiary alcohols |



Sandmeyer's Reaction

$\text{Cu}_2\text{Cl}_2/\text{HCl}$

$\text{Cu}_2\text{Br}_2/\text{HBr}$

CuCN/HCN

Gatterman's Reaction

Cu/HCl

Cu/HBr

- It is also used for preparing organocuprates (Gilman reagents)

Organic Chemistry Reagent Study Guide

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