

JASPERSE CHEM 360 TEST 3

VERSION 1

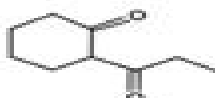
Ch 18 Ketones and Aldehydes

Ch 22 Additions and Condensations of Enols and Enolate Ions

1. Provide the Name for the Following (6 points)



trans-7-methoxy-4-hepten-3-one
(or E) (E)-7-methoxyhept-4-en-3-one



2-propanoylcyclohexanone

2. Of the following structures,

a. Which will be "completely" (>98%) deprotonated by LDA (LiN-*i*Pr₂)? (2 points)

A, B, C

b. Which will be "completely" (>98%) deprotonated by NaOH? (2 points)

C only



3. An unknown **X** has formula C₅H₈O. It gives 1) an orange precipitate upon treatment with 2,4-dinitrophenylhydrazine (2,4-DNP) and it gives 2) a silver mirror upon treatment with Tollen's reagent [Ag(NH₃)₂OH]. 3) It does not react with Br₂ in dichloromethane solvent. 4) Included in the ¹H NMR (incomplete) is a 6H doublet at 1.2 ppm. What is **X**? (4 points)

EU=1
C=O
aldehyde

NMR ⇒ isopropyl



4. Rank the rate of decarboxylation (loss of CO₂) for the following molecules upon heating, with 1 being highest, 2 being next, and 3 being not at all. [Hint: Two out of the three will react, one will not, so you should be able to identify the unreactive isomer. To compare the reactivity of the two reactive isomers, the phenyl substituent impacts the relative stabilities in the key step of the mechanism.] (2 points)



2



1



3

not 1,3 relationship

End stability:



normal



stabilized
by conjugation

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