



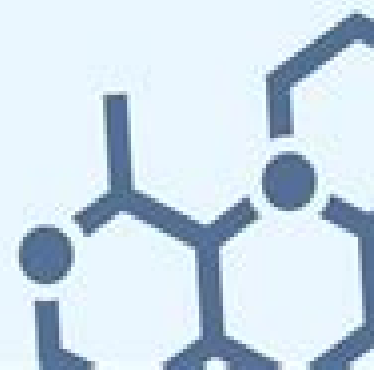
# Balancing Chemical Equation

(CH<sub>4</sub>) is:  $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

The number of carbon atoms is 1 on both sides of the equation, but the number of hydrogen atoms is 4 on the left side and 2 on the right side.

To balance this equation, you would need to place a coefficient of 2 in front of the H<sub>2</sub>O on the right side of the equation.

The balanced equation would be:  $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$



# Mcq On Balancing Chemical Equation

**Basic Concepts of Chemistry and MCQ  
for NEET**



## **Mcq On Balancing Chemical Equation:**

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