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Graham Stoakes,2025-04-08 Electric Vehicles Explained is your easy to follow companion on the journey to owning and driving an electric car. This comprehensive guide is designed with potential new and current electric car drivers in mind, offering clear explanations, practical tips, and insightful advice to make your transition to electric smooth and enjoyable. In this book, you will discover Why Go Electric. Uncover the numerous benefits of switching to an electric vehicle from environmental advantages to long term cost savings. Choosing the Right EV for You. Navigate the exciting world of electric cars and find the perfect match for your lifestyle and needs. Easy Charging Basics. Learn the ins and outs of charging your EV, including types of chargers, where to find them, and how to ensure you always have enough power. Driving an EV Tips for Beginners. Gain confidence behind the wheel with practical driving tips tailored specifically for electric vehicles. Navigating Range Anxiety. Say goodbye to range anxiety with strategies and tools to manage your vehicle's range effectively. Maintenance Made Simple. Discover how maintaining an EV is different from traditional cars and enjoy the simplicity and ease of EV upkeep. Building Confidence as an EV Driver. Boost your confidence on the road with advice on building your skills and knowledge as an electric vehicle driver. Electric Vehicles Explained combines straightforward descriptions with enough technical detail to satisfy inquisitive minds. Whether you're a tech savvy enthusiast eager to dive into the mechanics or someone looking for practical, no-nonsense advice, this book has something for you. Embrace the future of driving with confidence and excitement.

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web choose 1 answer the final temperature is closer to t 1 than to t 2 a the final temperature is closer to t 1 than to t 2 the final temperature is exactly halfway between t 1 and t 2 b the final temperature is exactly halfway between t 1 and t 2 the final temperature is closer to t 2 than to t 1 c

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web 6 a 218 g sample of steam at 121°C is cooled to ice at 14°C find the change in heat content of the system 7 in going from ice at 34°C to steam at 138°C a sample of water absorbs 1.41 x 105 J find the mass of the sample answers 1.268 x 104 J 2.623 x 104 J 3.171 x 104 J 4.111 x 105 J 5.211 x 105 J 6

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