



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the ions involved in the compound)



Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.
5. If the compound is an acid, use the prefix H- and the name of the anion.
6. If the compound is a base, use the name of the cation and hydroxide.

Binary Compounds

First Element	Anion
H	hydrogen
C	carbon
N	nitrogen
O	oxygen
F	fluorine
Cl	chlorine
S	sulfur
Br	bromine
I	iodine
P	phosphorus
As	arsenic
Sb	antimony
Bi	bismuth

Common Ions and Their Names

Formula	Name
Na ⁺	Sodium
K ⁺	Potassium
Ca ²⁺	Calcium
Mg ²⁺	Magnesium
Al ³⁺	Aluminum
Fe ²⁺	Iron (II)
Fe ³⁺	Iron (III)
Cu ²⁺	Copper (II)
Cu ⁺	Copper (I)
Zn ²⁺	Zinc
Ag ⁺	Silver
Li ⁺	Lithium
Rb ⁺	Rubidium
Cs ⁺	Cesium
Ba ²⁺	Barium
Sr ²⁺	Strontium
Pb ²⁺	Lead (II)
Pb ⁴⁺	Lead (IV)

Naming Compounds with a Central and Peripheral



Sodium Chloride

Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.

Naming Acids

Formula	Central Element	Formula	Central Element
HCl	Hydrogen	HNO ₃	Nitrogen
HBr	Bromine	HNO ₂	Nitrogen
HI	Iodine	H ₂ SO ₄	Sulfur
H ₂ SO ₄	Sulfur	H ₂ SO ₃	Sulfur
H ₂ SO ₃	Sulfur	H ₂ CO ₃	Carbon
H ₂ CO ₃	Carbon	H ₂ SiO ₃	Silicon
H ₂ SiO ₃	Silicon	H ₂ SiO ₂	Silicon

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Table salt	NaOH	Sodium hydroxide
KCl	Saline	Ca(OH) ₂	Slaked lime
Na ₂ CO ₃	Soda ash	CaCO ₃	Chalk
NaHCO ₃	Baking soda	Na ₂ SO ₄	Glauber's salt

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Wenbin Ji



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New International Dictionary, 1920 **An Introduction to Chemical Nomenclature** R. S. Cahn, 2013-12-19

Inorganic Chemical Nomenclature Lloyd McKinley, 1947 *An Introduction to Chemical Nomenclature* Robert Sidney Cahn, 1959 **Naming Chemical Compounds**, 2000 *Chemical Nomenclature* K. J. Thurlow, 1998 A general introduction to forms of chemical nomenclature dealing with systematic and trivial names Chapters are included on specialized naming systems for polymers and natural products and on the role of computers and the quest to find a quick and accurate naming program *The Etymology of Chemical Names*, 2019-10-08 Etymology of Chemical Names gives an overview of the development of the current chemical nomenclature tracing its sources and changing rules as chemistry progressed over the years This book is devoted to provide a coherent picture how the trivial and systematic names shall be used and how the current IUPAC rules help to reconcile the conflicting demands **The Naming and Indexing of Chemical Compounds** Chemical Abstracts (Periodical), 1962 *Chemical Nomenclature* American Chemical Society. Division of Chemical Literature, 1953 **The Naming and Indexing of Chemical Compounds** Chemical abstracts, 1962

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