

Naming Hydrocarbons

Steps for Naming Hydrocarbons	Example A	Example B
1 Examine the compound.	$ \begin{array}{c} \text{H} & \text{H} \\ & \\ \text{H} - \text{C} \equiv \text{C} - \text{C} - \text{C} - \text{H} \\ & \\ \text{H} & \text{H} \end{array} $	$ \begin{array}{c} \text{CH}_2 & & \\ & \diagup & \diagdown \\ & \text{CH}_2 & \text{CH}_2 \\ & & \\ \text{CH}_2 & & \text{CH}_2 \\ & \diagdown & \diagup \\ & \text{CH}_2 & \end{array} $
2 Count the number of carbon atoms in the longest continuous chain.	There are 4 carbon atoms in the longest chain.	There are 6 carbon atoms in the longest chain.
3 Determine the root name of the hydrocarbon using Table 2.	The root name is <i>but-</i> .	The root name is <i>hex-</i> .
4 Determine the type of bonds in the hydrocarbon, then use Table 3 to find the suffix.	There is a triple bond, so add the suffix <i>-yne</i> .	There are only single bonds, so add the suffix <i>-ane</i> .
5 Put the root and suffix together to name the hydrocarbon.	Combining the root and suffix gives the name <i>butyne</i> .	Combining the root and suffix gives the name <i>hexane</i> .
6 If the hydrocarbon is a ring, add <i>cyclo-</i> to the beginning of the name.	No prefix is needed because the structure is not a ring. The name of the hydrocarbon is <i>butyne</i> .	The structure is a ring so the prefix <i>cyclo-</i> is added to the name. The name of the hydrocarbon is <i>cyclohexane</i> .

Naming And Drawing Hydrocarbons Study Guide

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