

Chemistry Study Guide

The image shows a standard periodic table of elements. At the top center, it is titled "Periodic Table of the Elements". Below the title, there is a legend box with the heading "Element Symbols" and a list of elements with their symbols: H (Hydrogen), He (Helium), Li (Lithium), Be (Beryllium), B (Boron), C (Carbon), N (Nitrogen), O (Oxygen), F (Fluorine), Ne (Neon), Na (Sodium), Mg (Magnesium), Al (Aluminum), Si (Silicon), P (Phosphorus), S (Sulfur), Cl (Chlorine), Ar (Argon), K (Potassium), Ca (Calcium), Sc (Scandium), Ti (Titanium), V (Vanadium), Cr (Chromium), Mn (Manganese), Fe (Iron), Co (Cobalt), Ni (Nickel), Cu (Copper), Zn (Zinc), Ga (Gallium), Ge (Germanium), As (Arsenic), Se (Selenium), Br (Bromine), Kr (Krypton), Rb (Rubidium), Sr (Strontium), Y (Yttrium), Zr (Zirconium), Nb (Niobium), Mo (Molybdenum), Tc (Technetium), Ru (Ruthenium), Rh (Rhodium), Pd (Palladium), Ag (Silver), Cd (Cadmium), In (Indium), Sn (Tin), Sb (Antimony), Te (Tellurium), I (Iodine), Xe (Xenon), Cs (Cesium), Ba (Barium), Hf (Hafnium), Ta (Tantalum), W (Tungsten), Re (Rhenium), Os (Osmium), Ir (Iridium), Pt (Platinum), Au (Gold), Hg (Mercury), Tl (Thallium), Pb (Lead), Bi (Bismuth), Po (Polonium), At (Astatine), Rn (Radon), Fr (Francium), Ra (Radium), Ac (Actinium), Th (Thorium), Pa (Protactinium), U (Uranium), Np (Neptunium), Pu (Plutonium), Am (Americium), Cm (Curium), Bk (Berkelium), Cf (Californium), Es (Einsteinium), Fm (Fermium), Md (Mendelevium), No (Nobelium), and Lr (Lawrencium). The periodic table itself is organized into rows and columns, with elements grouped by their chemical properties. The elements are color-coded: metals are in shades of blue and green, non-metals are in shades of yellow and orange, and noble gases are in shades of purple and pink. The lanthanide and actinide series are shown at the bottom of the table.

Objective:

To understand the structure of an atom, and the nature + properties of subatomic particles 🌟🌟

Matter:

Anything that has mass and occupies space 🧑🌎

Examples: air, gases, liquids, solids?

Atoms:

Atoms are the smallest, indivisible particle 🧪👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- **Outer & Inner part** → together these are the **subatomic particles**

 Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Odysseyware Chemistry Study Guide

Gilbert William Castellan



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