

How to Name Chemical Compounds

is it ionic? is an anion,
or does it contain a polyatomic ion (as provided by H⁺)?

Ionic compound

[cation's name] [anion's name]

main- element	<ul style="list-style-type: none"> take its element name! ex. Al^{3+}: aluminum if transition metal*, include charge in Roman numeral ex. Cr^{2+}: chromium (II) 	<ul style="list-style-type: none"> get element's name, chop end, add "-ide" ex. F^{-}: fluorine → fluoride ✓
poly- element	<ul style="list-style-type: none"> just use its name! ex. NH_4^{+}: ammonium 	<ul style="list-style-type: none"> just use its name! ex. CO_3^{2-}: carbonate

* exceptions (no Roman numerals):

silver ion: always Ag^{+}

zinc ion: always Zn^{2+}

cadmium ion: always Cd^{2+}

Molecular compound

is it HCO_3^{-} ?
Acid! then use nomenclature

otherwise ...

[#]-[1st element name] [#]-[2nd element root]-"ide"

↑ skip if "mono-" ↑ skip if 1st element is H

Numerical prefixes

- 1: mono-
- 2: di-
- 3: tri-
- 4: tetra-
- 5: penta-
- 6: hexa-
- 7: hepta-
- 8: octa-
- 9: nona-
- 10: deca-

for
first element: never
ex. N_2O_5 → dinitrogen
pentoxide

Acids

Ignore all H^{+} , and look at the name of the remaining anion.

Anion	X-ide	X-ate	X-ite
Acid	hydro-X-ic acid	X-ic acid	X-ous acid
	ex. Br^{-} bromide HBr hydrobromic acid	ex. $\text{C}_2\text{H}_3\text{O}_2^{-}$ acetate $\text{HC}_2\text{H}_3\text{O}_2$ acetic acid	ex. IO_3^{-} iodate HIO_3 iodic acid
	ex. CN^{-} cyanide HCN hydrocyanic acid	ex. PO_4^{3-} phosphate H_3PO_4 phosphoric acid	ex. SO_3^{2-} sulfite H_2SO_3 sulfurous acid

Naming Molecular Compounds Study Guide

Bernice G. Segal, Peter S. Shenkin



Naming Molecular Compounds Study Guide:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J.

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Chemistry Gilleland, Gloria Mumford, 1982 *Students Study Guide* Fredeen, Pearson, 2003-07 **Fundamentals of Chemistry, Study Guide** James E. Brady, John R. Holum, 1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs **Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1992 **Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson** Norman E. Griswold, Henry Fuller Holtzclaw, 1988

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Table of Contents Naming Molecular Compounds Study Guide

1. Understanding the eBook Naming Molecular Compounds Study Guide
 - The Rise of Digital Reading Naming Molecular Compounds Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Molecular Compounds Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Molecular Compounds Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Molecular Compounds Study Guide
 - Personalized Recommendations

- Naming Molecular Compounds Study Guide User Reviews and Ratings
- Naming Molecular Compounds Study Guide and Bestseller Lists
- 5. Accessing Naming Molecular Compounds Study Guide Free and Paid eBooks
 - Naming Molecular Compounds Study Guide Public Domain eBooks
 - Naming Molecular Compounds Study Guide eBook Subscription Services
 - Naming Molecular Compounds Study Guide Budget-Friendly Options
- 6. Navigating Naming Molecular Compounds Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming Molecular Compounds Study Guide Compatibility with Devices
 - Naming Molecular Compounds Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Molecular Compounds Study Guide
 - Highlighting and Note-Taking Naming Molecular Compounds Study Guide
 - Interactive Elements Naming Molecular Compounds Study Guide
- 8. Staying Engaged with Naming Molecular Compounds Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Molecular Compounds Study Guide
- 9. Balancing eBooks and Physical Books Naming Molecular Compounds Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Molecular Compounds Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Molecular Compounds Study Guide
 - Setting Reading Goals Naming Molecular Compounds Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Molecular Compounds Study Guide
 - Fact-Checking eBook Content of Naming Molecular Compounds Study Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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760 761 762 763 764 765 766 767 768 769 770 771 772 773 774 775 776 777 778 779 780 781 782 783 784 785 786 787 788 789 790 791 792 793 794 795 796 797 798 799 800 801 802 803 804 805 806 807 808 809 810 811 812 813 814 815 816 817 818 819 820 821 822 823 824 825 826 827 828 829 830 831 832 833 834 835 836 837 838 839 840 841 842 843 844 845 846 847 848 849 850 851 852 853 854 855 856 857 858 859 860 861 862 863 864 865 866 867 868 869 870 871 872 873 874 875 876 877 878 879 880 881 882 883 884 885 886 887 888 889 890 891 892 893 894 895 896 897 898 899 900 901 902 903 904 905 906 907 908 909 910 911 912 913 914 915 916 917 918 919 920 921 922 923 924 925 926 927 928 929 930 931 932 933 934 935 936 937 938 939 940 941 942 943 944 945 946 947 948 949 950 951 952 953 954 955 956 957 958 959 960 961 962 963 964 965 966 967 968 969 970 971 972 973 974 975 976 977 978 979 980 981 982 983 984 985 986 987 988 989 990 991 992 993 994 995 996 997 998 999 1000 1001 1002 1003 1004 1005 1006 1007 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