

How to Name Chemical Compounds

Is it ionic or covalent?
or does it contain a polyatomic ion followed by H⁺?

Ionic compound

[Cation's name] [Anion's name]

metals
- same as element name
ex. Al³⁺: aluminum

- if transition metal²⁺,
include charge in
Roman numeral
ex. Cu²⁺: copper (II)

- if element's name,
drop end, add "-ide"
ex. Cl⁻: chlorine

poly-
- just use the name!
ex. NH₄⁺: ammonium

- just use H for the⁺
ex. CO₃²⁻: carbonate

* exceptions (no Roman numeral):

silver ion: silver Ag⁺

gold ion: gold-3 Au³⁺

cadmium ion: cadmium Cd²⁺

Molecular compound

[H]-[1st element name] [H]-[2nd element root]-"ide"

↑ skip if "mono-"

↑ skip if 1st element is H

Stem prefixes

- 1: mono-
- 2: di-
- 3: tri-
- 4: tetra-
- 5: penta-
- 6: hexa-
- 7: hepta-
- 8: octa-
- 9: nona-
- 10: deca-



Acids

Drop all H⁺, and look at the name of the remaining anion.

Acid	Anion	Acid	Anion
Acid	hydrochloric acid	Acid	Acid
ex. HCl ⁻ chloride HCl hydrochloric acid	ex. Ca(OH) ₂ hydroxide Ca(OH) ₂ calcium hydroxide	ex. HClO ⁻ chlorate HClO hydrochloric acid	ex. HClO ⁻ chloride HClO hypochlorous acid

Naming Molecular Compounds Study Guide

Bernice G. Segal, Peter S. Shenkin

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