
Nace cip level 1 study guide Questions with Complete Verified Solutions 2024

Define Corrosion

The deterioration of a substance, usually a metal from a reaction with its environment

What is passivation?

A layer of oxides formed on the surface of a metal that provides corrosion protection.

What is an example of passivation?

Stainless Steel

What are the elements of a corrosion cell?

Anode, Cathode, Metallic Pathway, Electrolyte

Describe what happens at the anode.

The metal dissolves into the electrolyte

What is the function of the electrolyte?

Allows the passage of ions to connect the corrosion cell

What is the function of the metallic pathway?

Allows the flow of electrons to connect the corrosion cell

What is the galvanic series?

A listing of metals in order of reactivity (more or less noble) in seawater at 25 C

The general rules of galvanic corrosion are:

The less noble (or more reactive) metal when connected to a more noble (or less reactive metal) will corrode preferentially

Name the five most important factors that affect the rate of corrosion.

Humidity, oxygen, chemical salts, pollutants, temperature

General corrosion is:

Results in a relatively uniform loss of material over the entire surface. Results in a general thinning of the affected surface. Relatively easy to inspect. Does not cause catastrophic failures

Localized corrosion is:

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Deeana Coopride, Harry W. Jarnagan

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Materials Performance ,2004 **Working Time Statistics** David Marsden,Statistical Office of the European Communities,1984 **School Library Journal** ,2001 **The British National Bibliography** Arthur James Wells,2003
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