

How to Name Chemical Compounds

Is it ionic or covalent?
or does it contain a polyatomic ion followed by H⁺?

Ionic compound

[Cation's name] [Anion's name]

metals	<ul style="list-style-type: none"> same as element name ex. Al³⁺: aluminum if transition metal⁺, include charge in Roman numeral ex. Cu²⁺: copper (II) 	<ul style="list-style-type: none"> if element's name, drop end, add "-ide" ex. Cl⁻: chlorine
polyatomic ions	<ul style="list-style-type: none"> just use the name! ex. NH₄⁺: ammonium 	<ul style="list-style-type: none"> just use H₁ for the¹ ex. SO₄²⁻: sulfate

* exceptions (no Roman numeral):

silver ion: silver Ag⁺

gold ion: gold(1) Au⁺

cadmium ion: cadmium Cd²⁺

Molecular compound

Is it ionic?

[H]-[1st element name] [H]-[2nd element root]-"ide"

↑ skip if "mono-"

↑ skip if 1st element is H

Stem/ radical prefixes

- 1: mono-
- 2: di-
- 3: tri-
- 4: tetra-
- 5: penta-
- 6: hexa-
- 7: hepta-
- 8: octa-
- 9: nona-
- 10: deca-



Acids

Drop all H⁺, and look at the name of the remaining anion.

Acid	Anion	Acid	Anion
HCl	chloride	H ₂ SO ₄	sulfate
H ₂ CO ₃	carbonate	H ₂ SO ₃	sulfite

Naming Molecular Compounds Study Guide

**Norman E. Griswold, Henry Fuller
Holtzclaw**

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Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon: buckyballs, graphite, and diamond are illustrated at the left as is the molecule methane CH₄ from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement. *Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03

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