

11 • Organic Chemistry**Organic Nomenclature Practice**

Name / Functional Group	Formula	Structural Formula
<i>n</i> -butane	(1) C_4H_{10} (2) $CH_3CH_2CH_2CH_3$	$\begin{array}{c} H_3C - C - CH_3 \\ \quad \\ H_2 \quad H_2 \end{array}$
name & functional group <i>2</i> -butanol alcohol name only	(1) and (2) C_4H_9OH $CH_3CH_2CH_2OH$	$\begin{array}{c} H_3C - C - OH \\ \quad \\ H_2 \quad H \end{array}$
<i>trans</i> - <i>2</i> -butene	C_4H_8 $CH_3CH=CHCH_3$	$\begin{array}{c} H_3C - H - C = CH_3 \\ \quad \quad \quad \\ \quad \quad \quad H \end{array}$
name only <i>butyne</i>	(1) and (2) C_4H_6 $C \equiv C - CH_2CH_3$	$\begin{array}{c} H_3C - H - C \equiv C \\ \quad \quad \quad \\ \quad \quad \quad CH_3 \end{array}$
name only <i>1,3-butadiene</i>	(1) and (2) C_4H_6 $CH_2=CHCH=CH_2$	$\begin{array}{c} H_3C - H - C = CH_2 \\ \quad \quad \quad \\ \quad \quad \quad H \end{array}$
name only <i>cyclopentane</i>	(1) only C_5H_{10}	$\begin{array}{c} H_2 \quad H_2 \\ \quad \\ C - C \\ \quad \\ H_2C - C - CH_2 \\ \quad \\ H_2 \quad H_2 \end{array}$
name only <i>cis</i> - <i>2</i> -butene	(1) and (2) C_4H_8 $CH_3CH=CCH_3$	$\begin{array}{c} H_3C - C = CH_3 \\ \quad \quad \quad \\ H \quad H \end{array}$
name only <i>trans</i> - <i>2</i> -butene	(1) and (2) C_4H_8 $CH_3C=CHCH_3$	$\begin{array}{c} H_3C - H - C = CH_3 \\ \quad \quad \quad \\ \quad \quad \quad H \end{array}$
name & functional group <i>methyl ethyl ether</i>	(1) and (2) C_3H_8O $CH_3CH_2OCH_3$	$\begin{array}{c} H_3C - C - O - CH_3 \\ \quad \\ H_2 \quad O \\ \quad \\ \quad \quad CH_3 \end{array}$

Naming Organic Compounds Answers

JL Elias

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