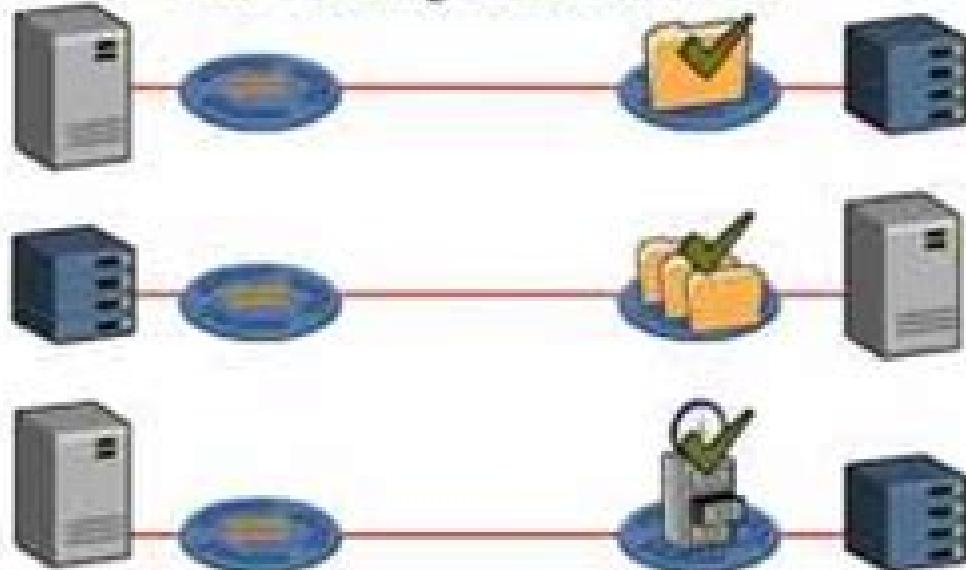


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## Differences between FTP and NDM

# Ndm Connect Direct Manual

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web 1 if the pe of the reactants is 430kj and the pe of the products is 190kj what is the  $\Delta H$  2 if the pe of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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