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web jul 12 2019 when we neutralize a weak acid with a strong base we get a salt that contains the conjugate base of the weak acid this conjugate base is usually a weak base for example sodium acetate NaCH_3CO_2 is a salt formed by the reaction of the weak acid acetic acid with the strong base sodium hydroxide

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web summary an arrhenius acid produces h and an arrhenius base produces oh in aqueous solutions acids taste sour may sting and neutralize bases bases taste bitter feel slippery and neutralize acids naming acids binary acids contain a single anion hnx to name hydro anion with ic end acid

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web definition of acids and bases identifying them along with the conjugate acid and conjugate base ph of strong and weak acids and bases the ph and poh relationship ka and kb pka and pka correlation the acid base properties of salts identifying the acidity of the salt based on the acid and the base it is prepared from and calculating the