

Atoms and Isotopes

An atom has a small central _____, which contains subatomic particles called _____ and _____. This is then surrounded by orbiting _____.

Complete the following table with the mass and charge of these subatomic particles.

	Relative Mass compared to a proton	Relative Charge
Proton		
Neutron		
Electron		

What is an atom called if it loses or gains electrons?

How does an atom become an isotope?

Label the numbers on the following element, taken from the Periodic Table.

number

Radon
→ 86

number

Tells us how many _____

P6 – Radioactivity – Triple Physics R

	Change in the Nucleus	Particle emitted and charge
α decay		
β decay		
γ decay		

Nuclear Equations

When a nucleus emits an alpha particle the atomic number goes down by _____ and the mass number goes down by _____.

When a nucleus emits a beta particle the atomic number goes up by _____ and the mass number stays the same.

What is the change in the atomic number and mass number when a nucleus emits gamma radiation? _____

Detecting Radiation

A teacher demonstrates a Geiger-Müller tube. It clicks when radiation enters the tube. The teacher looks at his results:

Radioactive source

Which source A, B or C emits Gamma radiation?

Half-Life

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KJ Lindholm-Leary



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