

13.1

THE NATURE OF GASES

Section Review

Objectives

1. describe the characteristics of gases and their physical properties
2. distinguish between an ideal and a real gas
3. define the macroscopic and kinetic properties and relationships of gases

Vocabulary

- | | | |
|---------------|-------------------------|------------------|
| a. ideal gas | b. pressure | c. pressure |
| d. atmosphere | e. atmospheric pressure | f. absolute zero |
| g. properties | h. volume | |

Part A Completion

Use the vocabulary terms in parentheses to complete each of the sentences. Write the complete sentence on the lines below.

1. Gases have three distinct a. properties that describe how they behave when the b. volume and c. pressure are changed. d. Temperature is also a property of gases. e. Atmospheric pressure is the pressure exerted by the f. atmosphere on the g. surface of the earth. h. Volume is the space that a gas occupies. i. Pressure is the force exerted by a gas on the j. surface of a container. k. Temperature is a measure of the l. average kinetic energy of the particles in a substance. m. Atmospheric pressure is the pressure exerted by the n. atmosphere on the o. surface of the earth. p. Volume is the space that a gas occupies. q. Pressure is the force exerted by a gas on the r. surface of a container. s. Temperature is a measure of the t. average kinetic energy of the particles in a substance.

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Part B True-False

Write T for true and F for false on the lines below. (If you are unsure, write U.)

- | | |
|-------------|--|
| <u> </u> | 1. Gases have three distinct properties. |
| <u> </u> | 2. Gases have a fixed volume. |

Nature Of Gases Section Review Answer Key

Jiyuan Zhang



Nature Of Gases Section Review Answer Key:

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