

NATURAL SCIENCES

Gr. 9 - LESSON PLAN - MATTER & MATERIALS

| TOPIC: COMPOUNDS | | |
|---|--|--|
| Sub-topic: Names of Compounds | | Duration: 1 Hour |
| CONTENT & CONCEPTS (CAPS p63) | | |
| KEY CONCEPTS | | |
| <ol style="list-style-type: none"> 1. Chemical reaction 2. Chemical formula 3. Names of compounds according to their elements. | | |
| Specific Aims: | Specific Aim 1: 'Doing Science' Specific Aim 2: 'Knowing the subject content and making connections' Specific Aim 3: 'Understanding the uses of Science' | <input checked="" type="checkbox"/> X <input checked="" type="checkbox"/> X |
| LESSON OBJECTIVES: | | |
| <ol style="list-style-type: none"> 1. Learners will be able to write the names of compounds as indicated on page 63 of the CAPS Document. 2. Learners will be able to make models of different compounds. | | |
| RESOURCES REQUIRED: | | |
| <ul style="list-style-type: none"> • Periodic Table Poster • Siyavula Sasol Izolilo (Learner book) page 163 | | |
| TEACHING & LEARNING ACTIVITY: | | |
| <p>Introduction.</p> <p>Briefly review and revise concepts dealt with in Grade 8, focusing on compounds.</p> <p>As each element has a unique name for example:</p> <ul style="list-style-type: none"> - Hydrogen - Lithium <p>Remember the first 20 elements learnt in grade 8. How did you memorize them? Do you remember Mnemonics...</p> <p>He - H; He - He; Likes - Li; Because - Be; Boys - B; Can - C; Not - N; Operate - O; Fireplaces - F;</p> <p>New - Ne; Nation - Na; Might - Mg; Also - Al; Sign - Si; Peace - P; Security - S; Clause - Cl;</p> <p>A - Ar; King - K; Can - Ca</p> <p>LEARNERS MAY RECITE THIS MNEMONIC AS A WAY OF MEMORIZING THE FIRST 20 ELEMENTS.</p> <p>The teacher provides the following explanation to the learners</p> <p>Each compound has a unique name. How is a compound formed?</p> <ul style="list-style-type: none"> - A compound is formed when TWO or more different elements combine chemically. - Each atom in a compound are held by chemical bonds - Compounds are formed during chemical reaction (when elements/ compounds combine to form new substances) - Each compound can be written as a formula, for example, H_2O; $NaCl$; $CuSO_4$ <p>What is a formula?</p> <ul style="list-style-type: none"> - This is a notation that shows which elements the compound is made up of. - It indicates the total number of atoms in one unit of the compound - It indicates the ratio in which atoms of different elements are bonded to each other - It gives the total number of atoms in one unit of a compound. <p>For example, the formula for Copper Sulphate is $CuSO_4$.</p> | | |

Natural Science Exemplar 2013 Grade 9

David Baud

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