

Question	
1	Q. What is the co-ordination no. of Simple Cubic crystal structure? A) 4 B) 6 C) 12
2	Q. Which one of the following is covalent crystal? A) dry ice B) rock salt C) ice
3	Q. Points which shows position of atoms in a crystal are called. A) lattice point B) lattice line C) lattice circle
4	Q. Crystal lattice is also known as. A) lattice triangle B) surface lattice C) lattice line D) lattice array
5	Q. The number of close neighbor in a body centered cubic lattice of identical spheres is. A) 6 B) 8 C) 12
6	Q. Which of the following is an example of ionic crystal solid? A) LiF B) Diamond C) Si
7	Q. Co-ordination number in Face Centered cubic crystal structure. A) 12 B) 8 C) 4
8	Q. On passing electricity through compound it acts as. A) Conductor B) Nonconductor C) Insulator
9	Q. Type of the cubic cell in the given structure. A) Hexagonal B) Body centered cubic C) Face centered cubic D) Simple cubic
10	Q. The number of atoms per unit cell of BCC is. A) 1 B) 2 C) 6 D) 7



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11	Q. The number of atoms present in molecule indicates (a) _____ A) Atomity B) Acidity C) Neutrality D) Alkalinity
12	Q. NaCl is an example of _____ A) Simple cubic B) Body centered cubic C) Face centered cubic
13	Q. The example of hexagonal closed packed structure is. A) Be B) Mg C) Ti
14	Q. The HCP structure has _____ layers. A) 3 B) 1 C) 4 D) 2
15	Q. On which factor conductance of metals is responsible? A) Ions B) delocalized C) atomic kernel D) number of atoms
16	Q. Amorphous solids are also referred as. A) hard B) liquid C) glassy
17	Q. Which of the following is ionic solid? A) Diamond B) Graphite C) sodium chloride
18	Q. Which of the following element becomes stable after losing an electron? A) Helium B) Iodine C) Sodium D) Oxygen
19	Q. Which of the following sentence is incorrect for covalent bond? A) Strength of covalent bond depends upon overlapping of atomic orbitals B) Covalent bond is not directional C) There is sharing of electrons between atoms bonded by covalent bond D) Covalent bond is formed between atoms having less difference in their electronegativity.
20	Q. Bonds formed by unequal sharing of electrons are called as. A) Non polar Covalent bond B) Polar ionic bond C) Non polar ionic bond

Multiple Choice Questions Of Solid State Chemistry

Joacim Rocklöv

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