

Investigation

1. Study of the \_\_\_\_\_ multi \_\_\_\_\_ relationships in a \_\_\_\_\_ situation.
2. Record of \_\_\_\_\_ balanced \_\_\_\_\_ equations.

2.  $40g + 40g \rightarrow 2 \text{ MgO}$

2. Magnesium reacting with oxygen to produce magnesium oxide.

The \_\_\_\_\_ in a \_\_\_\_\_ reaction \_\_\_\_\_ the \_\_\_\_\_ \_\_\_\_\_ from the \_\_\_\_\_ involved in the reaction.

3. Problem

Given a chemical equation, using the balanced equation, determine whether it is balanced.



Write down



2. How many atoms of Al, how many atoms of O, will be produced?

2.  $2 \text{ Al} + 3 \text{ O}_2 \rightarrow 2 \text{ Al}_2\text{O}_3$

# Note Taking Guide Episode 801 Answers Key

**Brendan G. Carr**



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