

Chemistry I (H) - Unit 02 (Pt. II - Nuclear Chemistry) Study Guide

1) Define, compare, and contrast radiation (pg. 122) to radioactivity (pg. 122).

2) Define AND give an example of a radioisotope (pg. 861).

3) Define nuclear reactions.

- Define nuclear equations (pg. 123/869) -
- Example:

4) Define and provide examples of the following forms of radioactive decay:

- alpha decay (pg. 867) -
 - alpha particle (pg. 123/862) -
 - Example Nuclear Equation:
- beta decay (pg. 867) -
 - beta particle (pg. 123/863) -
 - Example Nuclear Equation:
- gamma radiation (pg. 124) -
 - gamma ray (pg. 124/863) -
 - Example Nuclear Equation:
- Study the Table 24.2 from pg. 861 that we drew in our notes about the properties of decay (penetrating power, mass, etc.)

5) Define AND give an example of transmutation (pg. 865).

6) What do we mean by the words emit or emanate? Give an example.

7) Compare a stable nucleus to that of an unstable nucleus.

Nuclear Chemistry Study Guide Answers

Bernice G. Segal, Peter S. Shenkin

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Chemistry, Study Guide Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry chemical equilibrium and thermochemistry Worked examples now number over 300 and exercises over

1460 *Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy, 2011 **Basic Concepts of Chemistry, Study**

Guide Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry

Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems *General Chemistry Study*

Guide Sixth Edition Darrell D. Ebbing, 1999 **General Chemistry Workbook** Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives

grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material **General Chemistry, Study Guide** James E.

Brady, 1990-03-01 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems *Kaplan SAT Subject Test Chemistry 2015-2016* Kaplan Test Prep, 2015-03-03 Essential

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CliffsNotes AP Chemistry Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

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