1.2 (a)   What is a mineral?
     (b)   Describe the differences among the three main soil minerals.
     (c)   Which mineral group is most important for soils and why?

Solution 1.2

(a) Minerals are crystalline materials and make up the solids constituent of a soil. Minerals are classified according to chemical composition and structure.

(b) Kaolinite has a structure that consists of one silica sheet and one alumina sheet bonded together into a layer about 0.72 nm thick and stacked repeatedly. The layers are held together by hydrogen bonds. Tightly stacked layers result from numerous hydrogen bonds. Illite consists of repeated layers of one alumina sheet sandwiched by two silicate sheets. The layers, each of thickness 0.96 nm, are held together by potassium ions. Montmorillonite (smectite) has a structure similar to illite, but the layers are held together by weak van der Waals forces. Sodium smectite can absorb enough water to cause the particles to fully separate. Calcium smectites do not usually absorb enough water to cause particle separation because of their divalent cations. Montmorillonite is often called a swelling or expansive clay.

(c) Silicates, because this group of minerals is abundant on earth.

Muni Budhu Solution Manual

Minjie Lin

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