

## Chapter Test A

### Chapter: Chemical Equations and Reactions

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. You mix solution A with solution B in a beaker. Which of the following observations does *not* help you prove that a chemical reaction has occurred?
- The beaker becomes warm.
  - You see gas bubble out of solution.
  - Solution A, solution B, and the mixture are all yellow.
  - None of the above; all help prove that a chemical reaction has occurred.
- \_\_\_\_\_ 2. In the reaction described by the word equation  
 $\text{sodium} + \text{water} \rightarrow \text{sodium hydroxide} + \text{hydrogen gas}$
- sodium hydroxide is a product.
  - sodium is a product.
  - water is a reactant.
  - Both (a) and (c)
- \_\_\_\_\_ 3. Which of the following indicates that a chemical equation is balanced?
- The numbers of atoms of each element are the same on both sides of the equation.
  - All of the coefficients are the same.
  - The numbers of molecules on each side of the equation are equal.
  - The sums of the coefficients on each side of the equation are equal.
- \_\_\_\_\_ 4. From a complete and correctly written chemical equation, you can obtain the
- chemical formulas of the reactants and products.
  - relative amounts of the reactants and products.
  - physical states of the reactants and products.
  - All of the above
- \_\_\_\_\_ 5. To balance a chemical equation, you must
- adjust the subscripts.
  - adjust the number of products.
  - adjust the number of reactants.
  - adjust the coefficients.

# Modern Chemistry Teacher Resources Quizzes

**K Morrison**



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