

1. Please give an answer for the following question in Celsius. The problem requires:

$$T_{\text{initial}} = 27^\circ \text{C} \rightarrow T_{\text{final}} = 27^\circ \text{C}$$

The particle represented by N is:

A) He atom B) He ion C) He_2 molecule D) He_2^+ cation

2. Which isotopes will spontaneously decay and emit neutrinos with a charge of $-1/3$?

A) ^{15}O B) ^{16}O C) ^{17}O D) ^{18}O

3. Which of these particles has the greatest mass?

A) alpha B) beta C) neutrino D) proton

4. Please give an answer for the following question in Celsius. The problem requires:

$$\text{Proton} \rightarrow A \rightarrow \text{He}$$

The particle is represented by A :

A) alpha B) beta C) neutrino D) gamma

5. Alpha particles will form gamma rays when the:

A) mass, only B) charge, only C) both mass, and charge D) neither mass nor charge

6. Which statement best describes gamma radiation?

A) It has a mass of 1×10^{-24} charge of 0 . B) It has a mass of 0 charge of 0 . C) It has a mass of 0 charge of 0 . D) It has a mass of 0 charge of $-1/3$.

7. Which type of radiation is most similar to single-energy X-rays?

A) alpha B) beta C) gamma D) neutrino

8. Please use the initial masses of atoms that you gave. You are exactly 25,000 atoms of an isotope of ^{16}O (8-plets example of $\text{O} \times 14$ isotope subisotopes):

A) 3.26×10^{-23} g B) 5.01×10^{-23} g C) 1.1×10^{-23} g D) 1.7×10^{-23} g

9. Which isotopes are most commonly used for the detection of the movement of organic materials?

A) ^{35}Cl B) ^{37}Cl C) ^{39}Cl D) $^{37}\text{Cl}^+$

10. Look at Reference Table 3. What fraction of a radioactive ^{137}Cs sample would remain unchanged after 500 years?

A) $\frac{1}{2}$ B) $\frac{1}{4}$ C) $\frac{1}{8}$ D) $\frac{1}{16}$

11. Which is the healthiest used decay mode of Ra-226 ?

A) α -ray used alpha decay B) β -ray used beta decay C) γ -ray used alpha decay D) β -ray used beta decay

12. After 100 days, 2 milligrams of an α -radioactive isotope will be reduced to a sample containing neutrons. Which is the half-life of this element?

A) 5 days B) 12 days C) 100 days D) 400 days

13. Which reaction converts one atom of one element to one atom of another element?

A) ionization B) polymerization C) isotope fission D) isomerization

14. Radiation systems will be used in radiation therapy treatment. Could an undergo beta decay? This type of radiation reactions is called:

A) natural ionization B) artificial ionization C) nuclear fission D) nuclear fusion

15. In which reactions is energy converted to energy by the presence of "fission"?

A) $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$ B) $\text{H}_2\text{O} \rightarrow \text{H}_2$ C) $\text{H}_2\text{O} \rightarrow \text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O}$ D) $\text{H}_2 + \text{D}_2 \rightarrow \text{H}_2\text{O}$

16. Which statement best shows the relationship in a fission reaction?

A) Heavy nuclei split into lighter nuclei. B) Light nuclei fission into heavier nuclei. C) Energy is released and has visible effects on humans. D) Energy is absorbed and causes visible effects on humans.

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Puja Mehta

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